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METAMORPHIC RECORD IN METALHERZOLITE POCKETS WITHIN THE VIRSINI METAHARZBURGITE FROM THE KECHROS HP METAMORPHIC COMPLEX IN EASTERN RHODOPE, GREECE

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Abstract

The Virsini antigorite serpentinite from the Kechros HP metamorphic complex in eastern Rhodope is a serpentinized harzburgite with pockets of lherzolitic composition. In the metalherzolites the mantle assemblage is Ol-1+Opx+Cpx-1+Cr-Spl. The metamorphic assemblage is Atg+Ol-2+Cpx-2+Ol-3+Chl+Tr+Cr-Mag. Antigorite and Cr-magnetite formed in an early stage of the metamorphic event. With increase in metamorphic grade olivine (Ol-2) was formed at the expense of antigorite and with further increase tremolite, olivine (Ol-3) and chlorite were formed at the expense of clinopyroxene and antigorite. P-T conditions of ≈ 1.7 GPa/ ≈ 570 °C are recorded in neighbouring eclogites. For the temperature of 570 °C the reaction $Di+Atg \rightarrow Ol+Tr+H_2O$ occurs at pressure of ~ 1.2 GPa suggesting that the stability field of the mineral assemblage Atg+Ol+Tr in the metalherzolite is crossed during decompression.

Key words: Antigorite serpentinite, high pressure, Kechros complex, Rhodope.

Περίληψη

Ο αντιγοριτικός σερπεντινίτης της Βυρσίνης (Α. Ροδόπη) ανήκει στην υψηλών πιέσεων μεταμορφική ενότητα του Κέχρου. Πρόκειται για έναν σερπεντινιωμένο χαρτσβουργίτη με θύλακες λερζολιθικής σύστασης. Οι μεταλερζόλιθοι χαρακτηρίζονται από τη μανδυακή παραγένεση Ol-1+Opx+Cpx-1+Cr-Spl. Κατά τη μεταμόρφωση έχει σχηματιστεί το ορυκτολογικό άθροισμα Atg+Ol-2+Cpx-2+Ol-3+Chl+Tr+Cr-Mag. Ο αντιγορίτης και ο χρωμιούχος μαγνητίτης αναπτύσσονται κατά τα πρώτα στάδια της μεταμόρφωσης. Με αύξηση του βαθμού μεταμόρφωσης αναπτύσσεται αρχικά ολιβίνης (Ol-2) αντικαθιστώντας αντιγορίτη και στη συνέχεια τρεμολίτης, ολιβίνης (Ol-3) και χλωρίτης αντικαθιστώντας κλινοπυρόζενο και αντιγορίτη. Οι συνθήκες P-T στους γειτονικούς εκλογίτες είναι ≈1.7 GPa/≈570 °C. Για θερμοκρασία 570 °C η αντίδραση Di+Atg→Ol+Tr+H₂O λαμβάνει χώρα κατά την αποσυμπίεση σε πίεση ~1.2 GPa.

Λέζεις κλειδιά: αντιγοριτικός σερπεντινίτης, μεταμόρφωση υψηλών πιέσεων, ενότητα Κέχρου.

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1. Introduction

Peridotite bodies occur in numerous high-pressure (HP) metamorphic terrains within orogens involving major continental collisions. They usually form lens-shaped masses, normally within metamorphosed continental crustal rocks, but also within rocks of oceanic affinity. The lenses range in size from small (ca. 1 m across) to enormous (>100 Km² in outcrop area). In general, peridotites in gneiss terrains record the emplacement of mantle material into the continental crust and reveal the composition of the upper mantle and the nature of mantle/crust interaction processes (Brueckner and Medaris, 2000). Peridotites in metamorphic terrains potentially provide excellent information about temperature regimes. Phase relations for peridotites of harzburgitic and lherzolitic bulk compositions have been used to decipher details of the metamorphic evolution of a terrain (i.e.Trommsdorff and Evans, 1974; Mposkos et al., 2010; Padron-Navarta et al., 2011).

The Rhodope Domain extends over large areas of northern Greece and southern Bulgaria. It incorporates several tectonic slivers of reprocessed pre-Alpine crust, including Variscan granitoids, post-Variscan oceanic and continental margin sediments as well as igneous rocks subsequently subjected to HP metamorphism(s) (for protolith- and metamorphic ages see references in Burg, 2011; and Liati et al., 2011).

Mposkos and Krohe (2000) and Krohe and Mposkos (2002) subdivided the Greek part of the Rhodope domain into several tectonometamorphic units. These segments differ in P-T conditions during their Alpine HP metamorphism, and timing of exhumation. In the eastern Greek Rhodope the upper tectonic unit (Kimi complex; Mposkos and Krohe, 2000; 2006, Figure 1), shows a late Jurassic high-temperature eclogite facies, followed by Cretaceous granulite and upper amphibolite facies metamorphisms. The lower tectonic unit (Kechros complex; Mposkos and Krohe, 2000) experienced a low temperature eclogite facies followed by albite-epidote amphibolite- to lower amphibolite facies metamorphism (Mposkos, 1989; Mposkos et al., 2012). Ultramafic rocks occur in both tectonic units.

In the Kimi complex a garnet-spinel bearing lherzolitic meta-peridotite which contains layers of spinel-garnet clinopyroxenites, indicates an early high temperature and (U)HP metamorphism followed by cooling at substantial depth associated with relatively little decompression (Mposkos and Krohe, 2006). P-T conditions of >3 GPa and 1235 °C were estimated from the assemblage Ol-Opx-Cpx-Spl-Grt (abbreviations after Whitney and Evans, 2010). In the clinopyroxenites, garnetand spinel exsolution lamellae in clinopyroxenes, indicate that cooling started already in the garnet pyroxenite stability field. Hydration reactions like Grt+Opx+W→Ol+Hbl and Opx+Cpx+Spl+W→Hbl+Ol indicate decompression and cooling to below 1.2-1.3 GPa at 650-750 ^oC (Mposkos and Krohe, 2006).

In the Kechros complex large bodies of antigorite serpentinites like those from Dadia and Virsini areas (Figure1) represent the mantle assemblage. However, mineral assemblages, textures and mineral chemistry data which record the metamorphic evolution of these ultramafic rocks have yet to be reported.

In this paper we provide textural relationships and mineral compositions of the primary and the metamorphic mineral phases from metalherzolite pockets occurring within the Virsini antigorite serpentinite (metaharzburgite) and compare the metamorphic P-T path of the metaperidotite with that of the surrounding crustal rocks.

2. Geological Setting

The Kechros Complex consists of orthogneisses, reworked migmatites (containing muscovite metapegmatite lenses), pelitic gneisses, high-alumina metapelites and scarce marbles. Large serpentinized peridotites are tectonically intercalated. Within orthogneisses, serpentinites and between serpentinites and underlying metapelites, boudins of eclogites, eclogite amphibolites and

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amphibolites occur. HP metamorphism at P-T conditions of 1.7-2.2 GPa and 570-620 ⁰C were reported from the eclogites (Mposkos et al., 2012). In the orthogneisses and metapelites the HP event is indicated by the presence of high-Si phengites (Mposkos, 1989). In metapelites staurolite formed by the reactions $Cld + Ph \rightarrow St + Chl + Ms + Qz + H_2O$ and $Cld + Ms + Qtz \rightarrow St + Bt + H_2O$ H₂O suggests nearly isothermal decompression from maximum pressures of 2.2 GPa to 0.4 GPa (Mposkos 1989; Mposkos and Liati, 1993), implying rapid uplift. Orthogneisses and metapegmatites have Variscan protolith ages, as is indicated by Rb-Sr age of 334 Ma of magmatic muscovite from metapegmatite; (Mposkos and Wawrzenitz, 1995) and U-Pb ages between 326-299 Ma of magmatic zircons from orthogneisses (Peytcheva and von Quadt, 1995; Cornelius 2008; Liati et al., 2011). An U-Pb SHRIMP age of magmatic zircons from a kyanite-eclogite yielding 255 Ma is interpreted as the age of the gabbroic protolith (Liati et al., 2011) indicating that the subsequent HP metamorphism is Alpine in age. The minimum age of HP metamorphism is limited by Rb-Sr and Ar/Ar white mica ages of 36-42 Ma from mylonitic orthogneisses (Wawrzenitz and Mposkos, 1997; Lips et al., 2000). Emplacement of the Kechros complex to shallow crustal levels and juxtaposition against the Kimi complex by extensional deformation between 36 and 32 Ma is constrained by Rb-Sr biotite ages of granodiorites and subvolcanic rocks that intruded the Kechros complex and the overlying Kimi complex (Del Moro et al. 1988).



Figure 1 - Tectonic subdivision of eastern Greek Rhodope in northern Greece modified after Mposkos (1989) and Krohe and Mposkos (2002).

3. Ultramafic Rocks from the Kechros Complex

In the Kechros complex, metaperidotites occur as lens-shaped bodies (2 to 20 km in length and 0.5 to 5 km in width) intercalated between orthogneisses and metapelites. They are commonly associated with amphibolites and amphibolitized eclogites. The antigorite serpentinites from Dadia and Virsini areas (Figure 1) are the largest metaperidotite bodies in the Greek Rhodope. These ultramafic bodies contain dunitic boudins with chromitite pods, boudins of gabbroic pegmatites, diabase dykes, rodingitic gabbros and trondhjemites subsequently affected by HP metamorphism (Iliadis, 2006: Mposkos et al., 2012).

The Virsini antigorite serpentinite occupies an area of approximately 5×23 km and has a maximum thickness of about 300 m. It is a serpentinized and foliated clinopyroxene-bearing harzburgite with pockets of lherzolitic composition. Foliation is the result of penetrative deformation at albite-

epidote amphibolite/amphibolite facies conditions. The foliation strikes NE-SW and is similar to that of the surrounded phengite orthogneisses and metapelites.

	SiO ₂	TiO ₂	Al ₂ O ₃	Cr ₂ O ₃	Fe ₂ O ₃	Mn O	MgO	NiO	CaO	Na ₂ O	K ₂ O
LZ-A1	40.35	0.01	1.35	0.42	8.45	0.13	39.57	0.31	1.54	< 0.01	< 0.01
LZ-K1	42.80	0.02	1.70	0.29	8.34	0.11	37.35	0.26	1.54	0.05	0.01
LOI	Norm	Ol	En	Di	CaTs	MgTs	Chr				
7.0	LZ-A1	72.51	18.96	5.02	1.58	1.47	0.67				
7.16	LZ-K1	63.51	28.03	3.95	1.98	1.84	0.47				

Table 1: Bulk rock chemical- and norm composition of two metalherzolite samples (LZ- A1
and LZ-K1) from the Virsini metaperidotite.

4. Petrography

The studied metalherzolites crop out as poorly foliated to massive lenses within the antigorite serpentinite. They are composed of Ol±Opx+Cpx+Atg+Chl+Tr±Cr-Spl+Mag. Relicts of the mantle assemblage olivine (Ol-1), orthopyroxene, clinopyroxene (Cpx-1), and Cr-spinel are still preserved within an antigorite matrix. Primary olivine (Ol-1) up to 3 mm in size shows undulatory extinction and deformation bands. It is replaced by antigorite at the grain boundaries and along deformation bands and fractures. Aggregates consisting of isometric olivine grains (Ol-2, 25-100 µm in size) overgrow Ol-1 and are in textural equilibrium with matrix antigorite and Cr-magnetite (Figure 2a). Orthopyroxene (up to 6 mm in size) contains inclusions of olivine and exsolution lamellae of clinopyroxene. Orthopyroxene is preserved only in lherzolite pockets which were affected by low degree of serpentinization. In lherzolite pockets affected by high degree of serpentinization orthopyroxene is completely replaced by antigorite forming bastite pseudomorphs. Primary clinopyroxene (Cpx-1, up to 4 mm in size) is marginally replaced by thin prismatic tremolite and olivine (Ol-3). Olivine-3 forms aggregates of isometric to longitudinal grains. Scanning electron microscope (SEM) images show that the primary clinopyroxene is replaced by olivine lamellae (Ol-3), diopside (Cpx-2), tremolite, antigorite, chlorite and magnetite (Figures 2b,c). The alternating diopside, olivine, antigorite and chlorite lamellae shown in Figures 2b, 2c suggest that the primary clinopyroxene contained exsolution lamellae of orthopyroxene which are completely replaced by antigorite; with increasing grade of metamorphism antigorite lamellae were partly replaced by lamellar olivine and chlorite (see metamorphic conditions). Clinopyroxene lamellae contain vermicular exsolutions of Cr-magnetite (Figure 2c). Antigorite participates with > 60 % in volume in the rock. It forms up to 0.5 mm large and 0.05 mm thick flakes replacing olivine and pyroxenes. In clinopyroxene porphyroclasts, the antigorite and chlorite flakes grow parallel to the prismatic crystal faces of the clinopyroxene, probably replacing former orthopyroxene exsolutions (Figures 2b,d). Tremolite (up to 0.8 mm, commonly 0.3-0.6 mm in size) is prismatic and occurs in textural equilibrium with Ol-3 and chlorite replacing clinopyroxene and antigorite (Figure 2e). Large tremolite crystals contain inclusions of diopside, antigorite and magnetite (Figure 2f). Primary Cr-spinel is still preserved in the lherzolite pockets affected by a low degree of serpentinization. In lherzolite pockets affected by a larger degree of serpentinization Cr-spinel is completely replaced by chlorite and Cr-magnetite. This Cr-magnetite contains inclusions of diopside and antigorite.

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Figure 2 - Metalherzolite LZ-A1: a) Photomicrograph of isometric olivive grains (Ol-2) with inclusions of Cr-magnetite (black inclusions) b) Back-Scattered Electron (BSE) image of primary clinopyroxene replaced by diopside (Cpx-2), olivine (Ol-3), tremolite (Tr), antigorite (Atg) and chlorite (Chl). c) BSE image of lamellar clinopyroxene (Cpx-1) containing tiny chromite exsolutions. (Ol-3)=lamellar olivine, white grains Cr-magnetite, Atg=antigorite. d) BSE image pf alternating diopside (Cpx-2), antigorite and chlorite lamellae. The chlorite lamellae are in contact with lamellar olivine (Ol-3). e) BSE image of a zoned clinopyroxene (Al₂O₃= 1.16 wt % core, Al₂O₃=0 rim) with Cr-magnetite inclusions. Diopside (Cpx-2) and antigorite inclusions and is marginally replaced by olivine (Ol-3). An antigorite flake is enclosed in diopside and in the overgrown olivine. f) BSE image of diopside (Cpx-2) and antigorite replaced by olivine (Ol-3), chlorite and tremolite. Tremolite contains inclusion of diopside (right edge of the picture). Antigorite, olivine and chlorite are lamellar in shape and oriented in diopside.

5. Mineral Chemistry

Two samples from the metalherzolites, LZ-K1 and LZ-A1, (major and minor element- and norm compositions are given in Table 1), were selected for analyses of the mineral compositions. In sample LZ-K1, which shows a low degree of the metamorphic overprint, the mantle assemblage Ol-1+Opx+Cpx-1+Cr-Spl dominates, whereas in sample LZ-A1, the metamorphic assemblage Atg+Ol-2+Tr+Chl+Ol-3+Cr-Mag dominates. Relicts of Ol-1 and Cpx-1 are preserved in both samples.

Table 2: Metal	lherzolite: F	Representative	microprobe	analyses of or	thopyroxene,
clinopyroxene,	olivine, anti	igorite, chlorite	e, tremolite,	Cr-spinel and	Cr-magnetite.

	Opx	Opx	Cpx-1	Cpx-1	Cpx-2	Cpx-2	Ol-1	Ol-2/3	Tr	Atg	Atg
SiO ₂	55.59	55.50	52.00	52.22	55.13	55.02	40.90	40.95	58.52	41.07	42.11
Al ₂ O ₃	3.13	2.36	4.15	4.04	-	-	-	-	-	2.84	1.23
Cr ₂ O ₃	0.86	0.73	1.04	0.95	-	0.19	-	-	-	0.64	-
FeOt	5.65	6.15	3.13	3.03	1.17	0.91	8.58	9.15	1.49	3.35	4.66
MnO	-	-	-	-	-	-	-	-	-	-	-
MgO	33.27	33.40	19.28	18.92	17.72	17.79	50.11	49.51	23.77	40.12	39.95
NiO	-	-	-	-	-	-	0.37	0.41	-	0.16	0.28
CaO	1.48	1.40	20.36	20.80	25.83	25.76	-	-	13.63	-	-
Total	99.97	99.54	99.97	99.96	99.85	99.68	99.96	100.02	97.42	88.18	88.23
Si	1.921	1.931	1.882	1.890	2.001	1.999	0.998	1.001	7.996	7.638	7.857
Al	0.127	0.097	0.177	0.173	-	-	-	-	-	0.622	0.270
Cr	0.025	0.020	0.030	0.027	-	0.005	-	-	-	0.094	-
Fe	0.163	0.180	0.095	0.092	0.036	0.028	0.175	0.183	0.170	0.521	0.722
Mn	-	-	-	-	-	-	-	-	-	-	-
Mg	1.714	1.732	1.040	1.021	0.959	0.963	1.822	1.803	4.842	11.115	11.111
Ni	-	-	-	-	-	-	0.007	0.008	-	0.024	0.042
Ca	0.050	0.052	0.790	0.807	1.004	1.003	-	-	1.996	-	-
	Atg	Atg	Chl	Chl	Chl		Spl-c	Spl-r	Mag	Mag	Mag
SiO ₂	Atg 40.78	Atg 41.77	Chl 30.64	Chl 34.81	Chl 33.52	Al ₂ O ₃	Spl-c 43.73	Spl-r 40.58	Mag -	Mag 1.17	Mag
SiO ₂ Al ₂ O ₃	Atg 40.78 3.80	Atg 41.77 1.37	Chl 30.64 18.37	Chl 34.81 11.46	Chl 33.52 12.13	Al ₂ O ₃ Cr ₂ O ₃	Spl-c 43.73 25.97	Spl-r 40.58 28.68	Mag - 9.66	Mag 1.17 15.59	Mag - 10.30
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃	Atg 40.78 3.80	Atg 41.77 1.37 0.67	Chl 30.64 18.37 2.49	Chl 34.81 11.46 2.36	Chl 33.52 12.13 3.78	$\begin{array}{c} Al_2O_3\\ Cr_2O_3\\ Fe_2O_3\end{array}$	Spl-c 43.73 25.97 0.11	Spl-r 40.58 28.68 0.19	Mag - 9.66 59.67	Mag 1.17 15.59 53.57	Mag - 10.30 59.85
$\begin{array}{c} SiO_2\\ Al_2O_3\\ Cr_2O_3\\ FeO_t \end{array}$	Atg 40.78 3.80 - 3.38	Atg 41.77 1.37 0.67 5.82	Chl 30.64 18.37 2.49 3.17	Chl 34.81 11.46 2.36 3.42	Chl 33.52 12.13 3.78 3.68	$\begin{array}{c} Al_2O_3\\ Cr_2O_3\\ Fe_2O_3\\ FeO \end{array}$	Spl-c 43.73 25.97 0.11 13.34	Spl-r 40.58 28.68 0.19 15.21	Mag - 9.66 59.67 28.52	Mag 1.17 15.59 53.57 25.74	Mag - 10.30 59.85 26.93
	Atg 40.78 3.80 - 3.38 -	Atg 41.77 1.37 0.67 5.82	Chl 30.64 18.37 2.49 3.17	Chl 34.81 11.46 2.36 3.42	Chl 33.52 12.13 3.78 3.68	$\begin{array}{c} Al_2O_3\\ Cr_2O_3\\ Fe_2O_3\\ FeO\\ MnO \end{array}$	Spl-c 43.73 25.97 0.11 13.34 0.36	Spl-r 40.58 28.68 0.19 15.21 0.36	Mag - 9.66 59.67 28.52 -	Mag 1.17 15.59 53.57 25.74 0.33	Mag - 10.30 59.85 26.93 0.68
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO _t MnO MgO	Atg 40.78 3.80 - 3.38 - 39.83	Atg 41.77 1.37 0.67 5.82 - 38.28	Chl 30.64 18.37 2.49 3.17 - 33.34	Chl 34.81 11.46 2.36 3.42 - 35.61	Chl 33.52 12.13 3.78 3.68 - 34.57	Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ FeO MnO MgO	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96	Mag - 9.66 59.67 28.52 - 1.16	Mag 1.17 15.59 53.57 25.74 0.33 3.13	Mag - 10.30 59.85 26.93 0.68 2.13
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO _t MnO MgO NiO	Atg 40.78 3.80 - 3.38 - 39.83 0.22	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12	Chl 30.64 18.37 2.49 3.17 - 33.34	Chl 34.81 11.46 2.36 3.42 - 35.61	Chl 33.52 12.13 3.78 3.68 - 34.57 -	Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ FeO MnO MgO NiO	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96	Mag - 9.66 59.67 28.52 - 1.16 1.01	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66	Mag - 10.30 59.85 26.93 0.68 2.13 0.40
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO _t MnO MgO NiO CaO	Atg 40.78 3.80 - 3.38 - 39.83 0.22 -	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12	Chl 30.64 18.37 2.49 3.17 - 33.34 -	Chl 34.81 11.46 2.36 3.42 - 35.61	Chl 33.52 12.13 3.78 3.68 - 34.57 -	Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ FeO MnO MgO NiO	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96	Mag - 9.66 59.67 28.52 - 1.16 1.01	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66	Mag - 10.30 59.85 26.93 0.68 2.13 0.40
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO _t MnO MgO NiO CaO Total	Atg 40.78 3.80 - 3.38 - 39.83 0.22 - 88.01	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12 - 88.06	Chl 30.64 18.37 2.49 3.17 - 33.34 - - 88.02	Chl 34.81 11.46 2.36 3.42 - 35.61 - 87.69	Chl 33.52 12.13 3.78 3.68 - 34.57 - - 87.69	Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ FeO MnO MgO NiO	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57 - 99.98	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96 - - 99.97	Mag - 9.66 59.67 28.52 - 1.16 1.01 99.87	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66 100.20	Mag 10.30 59.85 26.93 0.68 2.13 0.40 100.3
$\begin{array}{c} SiO_2\\ Al_2O_3\\ Cr_2O_3\\ FeO_t\\ MnO\\ MgO\\ NiO\\ CaO\\ Total \end{array}$	Atg 40.78 3.80 - 3.38 - 39.83 0.22 - 88.01	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12 - 88.06	Chl 30.64 18.37 2.49 3.17 - 33.34 - - 88.02	Chl 34.81 11.46 2.36 3.42 - 35.61 - 87.69	Chl 33.52 12.13 3.78 3.68 - 34.57 - - 87.69	Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ FeO MnO MgO NiO	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57 - 99.98	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96 - - 99.97	Mag - 9.66 59.67 28.52 - 1.16 1.01 99.87	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66 100.20	Mag 10.30 59.85 26.93 0.68 2.13 0.40 100.3
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO _t MnO MgO NiO CaO Total Si	Atg 40.78 3.80 - 3.38 - 39.83 0.22 - 88.01 7.579	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12 - 88.06 7.864	Chl 30.64 18.37 2.49 3.17 - 33.34 - 88.02 5.755	Chl 34.81 11.46 2.36 3.42 - 35.61 - - 87.69 6.558	Chl 33.52 12.13 3.78 3.68 - 34.57 - 87.69 6.359	Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ FeO MnO MgO NiO NiO	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57 - 99.98 1.428	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96 - - 99.97 1.354	Mag - 9.66 59.67 28.52 - 1.16 1.01 99.87 -	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66 100.20 0.051	Mag - 10.30 59.85 26.93 0.68 2.13 0.40 -
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO _t MnO MgO NiO CaO Total Si Al	Atg 40.78 3.80 - 39.83 0.22 - 88.01 7.579 0.832	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12 - 88.06 7.864 0.308	Chl 30.64 18.37 2.49 3.17 - 33.34 - - 88.02 5.755 4.064	Chl 34.81 11.46 2.36 3.42 - 35.61 - - 87.69 6.558 2.545	Chl 33.52 12.13 3.78 3.68 - 34.57 - - 87.69 6.359 2.712	Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ Fe ₀ MnO MgO NiO NiO Al Cr	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57 - 99.98 - 1.428 0.569	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96 - - 99.97 1.354 0.642	Mag - 9.66 59.67 28.52 - 1.16 1.01 - 99.87 - 0.291	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66 100.20 0.051 0.456	Mag - 10.30 59.85 26.93 0.68 2.13 0.40 - 100.3 - 0.306
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO _t MnO MgO NiO CaO Total Si Al Cr	Atg 40.78 3.80 - 3.38 - 39.83 0.22 - 88.01 7.579 0.832 -	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12 - 88.06 7.864 0.308 0.100	Chl 30.64 18.37 2.49 3.17 - 33.34 - - 88.02 5.755 4.064 0.369	Chl 34.81 11.46 2.36 3.42 - 35.61 - - 87.69 6.558 2.545 0.352	Chl 33.52 12.13 3.78 3.68 - 34.57 - - 87.69 6.359 2.712 0.568	Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ Fe ₀ MnO MgO NiO NiO Al Cr Fe ³⁺	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57 - 99.98 - 1.428 0.569 0.002 -	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96 - - 99.97 1.354 0.642 0.004	Mag - 9.66 59.67 28.52 - 1.16 1.01 - 99.87 - 0.291 1.709	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66 100.20 0.051 0.456 1.492	Mag - 10.30 59.85 26.93 0.68 2.13 0.40 - 100.3 - 0.306 1.694
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO _t MnO MgO NiO CaO Total Si Al Cr Fe	Atg 40.78 3.80 - 3.38 - 39.83 0.22 - 88.01 7.579 0.832 - 3.380	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12 - 88.06 7.864 0.308 0.100 0.917	Chl 30.64 18.37 2.49 3.17 - 33.34 - - 88.02 5.755 4.064 0.369 0.498	Chl 34.81 11.46 2.36 3.42 - 35.61 - - 87.69 6.558 2.545 0.352 0.538	Chl 33.52 12.13 3.78 3.68 - 34.57 - - 87.69 6.359 2.712 0.568 0.584	$\begin{array}{c} Al_{2}O_{3} \\ Cr_{2}O_{3} \\ Fe_{2}O_{3} \\ FeO \\ MnO \\ MgO \\ NiO \\ NiO \\ \end{array}$	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57 - 99.98 1.428 0.569 0.002 0.309	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96 - - 99.97 1.354 0.642 0.004 0.360	Mag - 9.66 59.67 28.52 - 1.16 1.01 - 99.87 - 0.291 1.709 0.903	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66 100.20 0.051 0.456 1.492 0.797	Mag - 10.30 59.85 26.93 0.68 2.13 0.40 - 100.3 - 0.306 1.694 0.847
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO _t MnO MgO NiO CaO Total Si Al Cr Fe Mn	Atg 40.78 3.80 - 3.38 - 39.83 0.22 - 88.01 7.579 0.832 - 3.380 -	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12 - 88.06 7.864 0.308 0.100 0.917 -	Chl 30.64 18.37 2.49 3.17 - 33.34 - - 88.02 5.755 4.064 0.369 0.498 -	Chl 34.81 11.46 2.36 3.42 - 35.61 - - 87.69 6.558 2.545 0.352 0.538 -	Chl 33.52 12.13 3.78 3.68 - 34.57 - - 87.69 6.359 2.712 0.568 0.584 -	$\begin{array}{c} Al_{2}O_{3}\\ Cr_{2}O_{3}\\ Fe_{2}O_{3}\\ FeO\\ MnO\\ MgO\\ NiO\\ NiO\\ \hline\\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ $	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57 - 99.98 1.428 0.569 0.002 0.309 0.008	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96 - 99.97 1.354 0.642 0.004 0.360 0.009	Mag - 9.66 59.67 28.52 - 1.16 1.01 - 99.87 - 0.291 1.709 0.903 -	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66 100.20 0.051 0.456 1.492 0.797 0.010	Mag - 10.30 59.85 26.93 0.68 2.13 0.40 - 100.3 - 0.306 1.694 0.847 0.022
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO ₁ MnO MgO NiO CaO Total Si Al Cr Fe Mn Mg	Atg 40.78 3.80 - 3.38 - 39.83 0.22 - 88.01 - 7.579 0.832 - 3.380 - 39.830	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12 - 88.06 7.864 0.308 0.100 0.917 - 10.742	Chl 30.64 18.37 2.49 3.17 - - 33.34 - - 88.02 5.755 4.064 0.369 0.498 - 9.355	Chl 34.81 11.46 2.36 3.42 - 35.61 - 87.69 6.558 2.545 0.352 0.352 0.352 0.352 - 10.000	Chl 33.52 12.13 3.78 3.68 - 34.57 - 87.69 6.359 2.712 0.568 0.584 - 9.777	$\begin{array}{c} Al_{2}O_{3}\\ Cr_{2}O_{3}\\ Fe_{2}O_{3}\\ FeO\\ MnO\\ MgO\\ NiO\\ \hline\\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ $	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57 - 99.98 1.428 0.569 0.002 0.309 0.008 0.685	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96 - 99.97 1.354 0.642 0.004 0.360 0.009 0.631	Mag - 9.66 59.67 28.52 - 1.16 1.01 - 99.87 - 0.291 1.709 0.903 - 0.066	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66 100.20 0.051 0.456 1.492 0.797 0.010 0.173	Mag
SiO ₂ Al ₂ O ₃ Cr ₂ O ₃ FeO _t MnO NiO CaO Total Si Al Cr Fe Mn Mg Ni	Atg 40.78 3.80 - 3.38 - 39.83 0.22 - 88.01 7.579 0.832 - 3.380 - 39.830 0.220	Atg 41.77 1.37 0.67 5.82 - 38.28 0.12 - 88.06 7.864 0.308 0.100 0.917 - 10.742 0.020	Chl 30.64 18.37 2.49 3.17 - - 33.34 - - 88.02 5.755 4.064 0.369 0.498 - - 9.355 -	Chl 34.81 11.46 2.36 3.42 - 35.61 - 87.69 6.558 2.545 0.352 0.352 0.352 - 10.000	Chl 33.52 12.13 3.78 3.68 - 34.57 - 87.69 6.359 2.712 0.568 0.584 - 9.777 -	$\begin{array}{c} Al_{2}O_{3}\\ Cr_{2}O_{3}\\ Fe_{2}O_{3}\\ FeO\\ MnO\\ MgO\\ NiO\\ \hline\\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ $	Spl-c 43.73 25.97 0.11 13.34 0.36 16.57 - 99.98 1.428 0.569 0.002 0.309 0.008 0.685	Spl-r 40.58 28.68 0.19 15.21 0.36 14.96 - - 99.97 1.354 0.642 0.004 0.360 0.009 0.631	Mag - 9.66 59.67 28.52 - 1.16 1.01 - 0.291 1.709 0.903 - 0.066 0.031	Mag 1.17 15.59 53.57 25.74 0.33 3.13 0.66 100.20 0.051 0.456 1.492 0.797 0.010 0.173 0.020	Mag 10.30 59.85 26.93 0.68 2.13 0.40 100.3 - 0.306 1.694 0.847 0.022 0.119 0.012

c=core, r=rim. Analyses were performed using a JEOL 6380 LV SEM equipped with energy dispersive system (EDS) at the School of Mining and Metallurgical Engineering, National Technical University of Athens. Representative mineral compositions are given in Table 2.

5.1. Mantle Assemblage:

The forsterite component in olivine (Ol-1) ranges between $Fo_{0.89}$ and $Fo_{0.91}$. In orthopyroxene the Al₂O₃ content ranges from 2.27 to 3.13 wt %, the Cr₂O₃ from 0.46 to 0.86 wt %, the CaO content

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from 1.27 to 1.48 wt % and the Mg/(Mg+Fe) ratio from 0.9 to 0.91. Mantle clinopyroxene (Cpx-1) is Cr-diopside. The Al₂O₃ content ranges from 1.73 to 4.15 wt %, the Cr₂O₃ from 0.48 to 1.92 wt % and the Mg/(Mg+Fe) ratio from 0.92 to 0.94. Lower Al₂O₃ contents show rim compositions in zoned ortho- and clinopyroxene grains. Chromite is Cr-spinel with Cr/(Cr+Al+Fe³⁺) ratio 0.28 in the core and 0.32 at the rim and Mg/(Mg+Fe²⁺) ratio 0.68 and 0.63 respectively. The decrease in Al and Mg content from the core to the rim is attributed to alteration and partial replacement of the Cr-spinel by chlorite during the metamorphic processes.

5.2. Metamorphic assemblage:

The forsterite component in Ol-2 and Ol-3 ranges between $Fo_{0.90}$ and $Fo_{0.91}$. No remarkable differences are observed in the forsterite component between mantle Ol-1 and metamorphic Ol-2 and Ol-3. Cpx-2 shows a higher Mg/(Mg+Fe) ratio (0.96-0.97) compared to that of Cpx-1 (0.92-0.94). In Cpx-2 the Al₂O₃ content is below the detection limit and the Cr₂O₃ content from 0 to 0.19 wt%. Antigorite contains appreciable amounts of aluminium. The Al₂O₃ content ranges from 0.5 to 4.28 wt %. Higher Al₂O₃ values are measured in antigorite flakes replacing Cpx-1. The Cr₂O₃ content ranges from 0 to 1.52 wt %. Chlorite is clinochlore with 5.75 to 6.70 Si atoms per formula unit (a.p.f.u.), 2.21 to 4.06 Al a.p.f.u. and a Mg/(Mg+Fe) ratio ranging from 0.93 to 0.95. The Cr₂O₃ content in chlorite ranges from 2.36 to 3.78 wt % and total iron, expressed as FeO, from 3.17 to 4.49 wt %. The Mg/(Mg+Fe) ratio in tremolite ranges between 0.95 to 0.98; the Al₂O₃ content is below the detection limit. Cr-magnetite is formed at the expense of primary spinel, olivine (Ol-1), and clinopyroxene within the stability field of antigorite during the serpentinization. The Cr-magnetite inclusions in Ol-2 (Figure 2b) indicate that it is formed at an early stage of the prograde path of metamorphism. The Cr₂O₃ content ranges from 9.21 to 14.95 wt %, the MgO from 0.23 to 1.45 wt % and the Al₂O₃ content from 0 to 1.17 wt %.

6. Discussion

Textural relationships, mineral assemblages and mineral compositions in the metalherzolite pockets indicate that the Kechros metaperidotite is affected by the HP metamorphic event recorded in the neighboring crustal rocks (Mposkos, 1989; Mposkos et al., 2011). The mantle assemblage was Ol-1+Opx+Cpx-1+Cr-Spl and the metamorphic assemblage is Atg+Ol-2+Cpx-2+Ol-3+Chl+Tr+Cr-Mag. Mantle olivine (Ol-1) and orthopyroxene are replaced by antigorite. Antigorite and Cr-magnetite are formed in an early stage of the metamorphic event as indicate antigorite and Cr-magnetite inclusions in olivine (Ol-2 and Ol-3, Figure 2a). Antigorite occurs also as inclusion in Cr-magnetite. In most metalherzolite pockets orthopyroxene is absent. It is completely consumed by the antigorite forming reaction:

$$Ol+Opx+W\rightarrow Atg$$
 (1)

With increasing metamorphic grade and by crossing the olivine plus antigorite stability field, Ol-2 is formed at the expense of antigorite. With further increase of the metamorphic grade, or during decompression, tremolite, olivine (Ol-3) and chlorite are formed at the expense of clinopyroxene (Cpx-1) and antigorite as suggested by the textural relationships (Figures 2b,e,f). Tremolite and olivine are formed according to the reaction:

$$Atg+Di \rightarrow Ol+Tr+H_2O$$
 (2)

Antigorite lamellae in clinopyroxene are probably formed replacing lamellae of enstatite exsolutions in the primary clinopyroxene (Figures 2b,d) according to reaction 1. Olivine lamellae in clinopyroxene (Figures. 2b,c) are formed replacing antigorite lamellae according to reaction 2.

Chlorite is also formed by the replacement of the clinopyroxene, in addition to tremolite and olivine-3 (Figure 2f). The Tschermak component of the primary clinopyroxene (Cpx-1) was consumed according to the reaction:

$$Atg+Cts \rightarrow Tr+Fo+Chl+W$$
 (3)

The Al_2O_3 content in antigorite (it ranges from 0.5 to 4.28 wt %) was also an aluminum source for the chlorite formation.

The maximum P-T conditions were lower than those constrained by the reaction:

$$Atg \rightarrow Ol+Tlc+H_2O$$
 (4)

since antigorite is still stable and the assemblage olivine+talc was not formed.



Figure 3 - Peak metamorphic PT conditions (adopted from neighboring eclogites, Mposkos et al., 2012, dotted line) and exhumation path of the Virsini metaperidotite from the Kechros complex in eastern Rhodope.

The metamorphic temperatures are constrained within the narrow field confined by the curves of the reactions 2 and 4 (Figure 3). This is also in accordance with the low Al-spinel component in the Cr-magnetite (up to ≈ 2 %) which suggests that Cr-magnetite is formed below the talc + olivine stability field (Evans and Frost, 1975). In talc-olivine rocks ferrit-chromites accommodate Al-spinel component up to 5 % and in chlorite-enstatite-olivine rocks up to 35 % (Evans and Frost, 1975; Mposkos et al., 2010). Boudins of amphibolitized eclogites and of omphacite-bearing meta-trondhjemites within the Virsini metaperidotite (Mposkos et al., 2012; Mposkos et al., 2013) indicate that the Kechros ultramafic rocks are metamorphosed at high pressures. For pressure of about 1.7 GPa (recorded in neighbouring eclogites) and assuming that the reaction 2 occurred at the prograde stage of metamorphism, the maximum temperature in the metaperidotite is constrained between 610 and 630 °C which is slightly higher than that obtained from the associated eclogites (≈ 570 °C at 1.7 GPa; Mposkos et al., 2012). At peak temperature of approximately 570 °C the curve of the reaction 2 is crossed during decompression. Assuming isothermal decompression as was recorded in the associated eclogites and metapelites the reaction curve was crossed at a pressure of ~ 1.2 GPa (Figure 3).

7. Conclusions

The Virsini metaperidotite from the Kechros HP metamorphic complex in eastern Rhodope is a sepentinized clinopyroxene-bearing harzburgite with pockets of lherzolitic composition. Dykes of amphibolitized eclogites and omphacite bearing metatrondjemites within the metaperidotite indicate that the metaperidotite underwent HP metamorphism similar to that recorded in the associated rocks of the continental crust assemblage.

In the metalherzolite pockets the mantle assemblage is Ol-1+Opx+Cpx-1+Cr-Spl. The metamorphic assemblage is Atg+Ol-2+Cpx-2+Ol-3+Chl+Tr+Cr-Mag. Antigorite and Cr-magnetite are formed in an early stage of the metamorphic event. Antigorite lamellae in clinopyroxene are probably formed replacing lamellae of "enstatite exsolutions" in the primary clinopyroxene. With increasing metamorphic grade and by crossing the olivine plus antigorite stability field olivine (Ol-2) is formed at the expense of antigorite. With further increase of metamorphic grade (or more probably during decompression), tremolite, olivine (Ol-3) and chlorite are formed at the expense of clinopyroxene and antigorite. Olivine lamellae in clinopyroxene replace antigorite- or former orthopyroxene lamellae in clinopyroxene. The aluminum source for the chlorite formation was the Al₂O₃ content in antigorite (up to 4.28 wt %) and the Tschermak component of the primary pyroxenes. The metamorphic temperatures are constrained within the stability field of the mineral assemblage Atg+Ol+Tr (Figure 3). At peak PT conditions (≈ 1.7 GPa/570 °C recorded in neighbouring eclogites) the curve of the reaction Di+Atg→Ol+Tr+H₂O is crossed during decompression.

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