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# AESTHETIC CHARACTERISTICS OF GREEK ORNAMENTAL STONES ASSOCIATED WITH MINERAL, GEOCHEMICAL AND STRUCTURAL PROPERTIES

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#### Abstract

Commercial marble are almost ready-to-use materials with a high economic value for the building-construction industry. The most important ornamental stones in Greece include limestone and marble with notable qualitative features and are considered products of high commercial value. In this research, seventy-three carbonate samples (limestones, dolomites and marbles) from all over Greece have been studied to determine their macroscopic and petrographic features and to investigate their mineral chemistry and geochemistry. Their colors vary from whitish to gray due to their content of calcite and dolomite, black due to bitumens, yellow due to the presence of iron oxides and clay, red due to iron oxides and mixed phases of calcite- siderite, green-brownish due to the presence of iron oxides or muscovite and chlorite, giving schistosity to the rock. Scattered calcite and dolomite veins, as well as the occurrence of other minerals such as barite, epidote and magnetite, contribute to specific features, thus making their appearance unique. The particular characteristics of each stone depend highly upon the type of minerals present, their composition, grain size and extent of aggregation, their geochemistry and structure. Keywords: Ornamental stones, marble, aesthetic features, mineral chemistry, geochemistry.

#### Περίληψη

Τα εμπορικά μάρμαρα αποτελούν υλικά που δεν απαιτούν ιδιαίτερη επεξεργασία με υψηλή οικονομική αξία για την κατασκευαστική βιομηχανία. Τα πιο σημαντικά διακοσμητικά πετρώματα στην Ελλάδα περιλαμβάνουν ασβεστόλιθους και μάρμαρα με ιδιαίτερα ποιοτικά χαρακτηριστικά και θεωρούνται προϊόντα ιδιαίτερης εμπορικής αζίας. Σε αυτή την εργασία μελετήθηκαν εβδομήντα τρία ανθρακικά δείγματα (ασβεστόλιθοι, δολομίτες και μάρμαρα) από όλη την Ελλάδα με σκοπό να καθοριστούν τα μακροσκοπικά και πετρογραφικά χαρακτηριστικά τους και να διερευνηθεί η ορυκτοχημική τους σύσταση και η γεωχημεία τους. Τα χρώματά τους ποικίλουν από λευκό έως τεφρό, εξαιτίας του περιεχομένου τους σε ασβεστίτη και δολομίτη, μαύρο σε βιτουμενιούχους ασβεστόλιθους, κίτρινο εξαιτίας της παρουσίας οζειδίων σιδήρου και αργιλικών ορυκτών, κόκκινο εξαιτίας των οζειδίων σιδήρου και μεικτών φάσεων ασβεστίτη και σιδηρίτη, καστανοπράσινο εξαιτίας της εμφάνισης οξειδίων σιδήρου μαζί με χλωρίτη και μοσχοβίτη, προσδίδοντας σχιστότητα στο πέτρωμα. Διάσπαρτες ασβεστιτικές και δολομιτικές φλέβες, όπως επίσης και εμφάνιση άλλων ορυκτών φάσεων όπως βαρύτης, επίδοτο και μαγνητίτης συμβάλλουν στη δημιουργία ιδιαίτερων χαρακτηριστικών που προσδίδουν μοναδική εμφάνιση. Τα χαρακτηριστικά γνωρίσματα κάθε πετρώματος εξαρτώνται σημαντικά από το είδος των ορυκτών, τη σύσταση, κοκκομετρία και κατανομή τους, τη γεωχημεία και δομή τους.

**Λέξεις κλειδιά:** Διακοσμητικά πετρώματα, μάρμαρα, αισθητικά χαρακτηριστικά, ορυκτοχημεία, γεωχημεία.

# 1. Introduction

The aesthetic characteristics of marbles depend on their color variability, as well as on their mineralogy and physical features (texture, grain size and other structural features). These parameters can change even within the same pit environment causing color differences in quarried marble blocks. Differences in material content and metamorphic processes influence the final colors of marbles, whereas impurities in the original rock mass also contribute in the different color varieties. Iron-oxides in original rock mass for example result in forming yellow, yellowish or reddish marble types. Manganese-oxides and carbonic impurities in original rock mass cause the formation of dark bluish, violet or black colored marbles. After metamorphism, marbles as rock masses have not stayed static in their locations and they have been influenced by all types of structural defects. Most of the defects in marble mass involve discontinuities, some of which are open, disturbing the continuity of the marble blocks and others are filled with loose or compact materials. Compact filling materials cause color changes and provide additional structural features affecting marble appearance. The marble aesthetic characteristics are directly linked with their commercial final market price, which is decided by marble users and interior designers. (Brilli *et al.*, 2015; Directory of Greek Ornamental and Structural Stones, 2015; Ferrini *et al.*, 2012; Munyanyiwa *et al.*, 1988; Papatrechas, 2011).

In general commercial marbles include many others non-carbonate rocks capable of taking a polish. These rocks such as limestones, travertines, dolomites, marbles, granites, basalts, schists, sandstones, etc. are mined for the construction industry, especially to be used as decorative stones. In Greece, marble industry is one of the most effective forces in economy. In this paper, we present the aesthetic features of characteristic ornamental marbles from many areas of Greece in relation to their petrographic, mineralogical, chemical, and geochemical data.

# 2. Sampling and analytical methods

In this study, 73 samples have been analysed for mineralogical and geochemical investigations. For this purpose, 17 samples were collected from Eastern Macedonia, 6 samples from Central Macedonia, 4 samples from Western Macedonia, 9 samples from Peloponnesus, 4 samples from Thessaly, 1 samples from North Aegean, 2 samples from South Aegean, 7 samples from Crete, 15 samples from Sterea Ellada, 2 samples from Western Greece and 6 samples from Epirus.

Mineralogic and petrographic parameters (i.e. mineral association, grain size and structure, texture of samples) were examined under polarized light microscope using thin sections as well as with scanning electron microscopy–energy dispersive spectroscopy (SEM–EDS), using a JEOL JSM 5600 scanning electron microscope, equipped with an automated energy dispersive analysis system ISIS 300 OXFORD, with the following operating conditions: 20 kV accelerating voltage, 0.5 nA beam current, 20 s time of measurement and 5 µm beam diameter.

Qualitative mineralogical composition of the marble samples was determined with X-ray powder diffraction (XRD), using a Philips X'Pert Panalytical X-ray diffractometer, operating with Cu radiation at 40 kV, 30 mA, 0.020° step size and 1.0 sec step time. The XRD patterns were evaluated using the DIFFRAC plus EVA software v.11 (Bruker-AXS, USA) based on the ICDD Powder Diffraction File

(2006). Major oxides and the trace elements analysis of all samples was carried out in the IGME laboratory. Samples were pulverized to <200 mesh in an agate mill, were digested with a mixture of HCl–HNO<sub>3</sub>–HF acids and were analyzed for a series of trace elements by Inductively Coupled Plasma-Atomic Emission Spectroscopy (ICP-MS) and for major elements by X-ray fluorescence (XRF).

### 3. Mineralogical and petrographic characterization of the samples

### 3.1 Macroscopic examination

Macroscopic investigations revealed that the marble samples can be grouped into eight types based on colo r features, crystal size, crystal boundaries and foliation. These groups are identified as (i) white, (ii) semi-w hite - light gray, (iii) beige-yellow, (iv) pink, (v) red, (vi) greenish, (vii) gray and (viii) blackish marbles. Th e ornamental marbles can be classified into three groups, based on their petrological characteristics . These groups are identified as **marbles** (AM\_1- 17, KM\_1-2, DM\_1-4, P\_1-2, TH\_1-4, NA\_1-2, STE\_1-8, K\_1-2), **limestones** (KM\_3-6, P\_3-6, BA\_1, P\_7, STE\_9-11, DS\_1-2, H\_1-6, K\_3-7) and **dolomites** (P\_8-9, STE\_12-15).

### 3.2 Microscopic and XRD examination

Microscopically, the marble groups display homeoblastic or heteroblastic mosaic texture, with vary ing crystal size. Larger crystal sizes appear in marbles from Macedonia, South Aegean and Sterea Ellada (cipolin marbles). The limestone-dolomite groups display a homeoblastic or heteroblastic te xture, with small crystal sizes.

Calcite and dolomite represent the major crystalline phases of the studied samples. Other minerals that participate in the mineralogical composition are quartz, muscovite, biotite, clinochlore, chlorit e, epidote, clay minerals (illite, kaolinite), magnetite, apatite, fluorite, sphalerite, etc. Calcite is the main mineral in samples AM\_5, AM\_8-11, AM\_13, KM\_1-6, DM\_1-4, P\_1-7, BA\_1, TH\_1-4, NA\_1-2, STE\_1-11, DS\_1-2, H\_1-6, K\_1-7, while dolomite is the main mineral in samples AM\_1-4, AM\_6-7, AM\_14-17, P\_8-9, STE\_12-15. Calcitic marbles mainly consist of 75-100% calcite and 0-20% dolomite, dolomitic marbles consist of 88-99% dolomite and 1-12% calcite, limestones consist of 85-100% calcite and 1% dolomite, and dolomites consist of 90-96% dolomite and 3-10% calcite. Calcite bearing rocks (both marbles and limestones) may also contain up to 15% micas.

The color presented by a marble, depends on its composition or its granulometry. White, semi-whit e, light gray and beige colors can be attributed either to pure calcite or dolomite. For example, the c alcitic marble NA\_1 consists of 100% calcite (Fig. 1.1), while the dolomitic marble AM\_3 consists mainly of dolomite (white crystalline mass), while calcite crystals are concentrated at few faint gra y areas (Fig. 1.2). At the semi-white calcite marble DM\_3, the white crystalline mass consists of ca lcite and the gray strips consist of dolomite crystals with smaller size (Fig. 1.3). In the white calciti c marble DM\_1, the white crystalline mass consists of calcite with granoblastic texture and the yell ow strips consist of smaller size calcite crystals mixed with clay minerals (Fig. 1.4). In the yellowwhite calcitic marble AM 15, the crystalline mass comprises of small dolomitic crystals and a net work of yellow-red veins mainly of iron and manganese oxides (Fig. 1.5). Beige-yellow colors usu ally indicate the presence either of clay minerals or iron oxides. Pink and red marbles mainly attrib ute their color to iron and manganese impurities in the mass of calcite or dolomite, as calcium or m agnesium are substituted by these elements. In the pink calcitic marble TH 2, the crystalline mass consists of pink calcitic crystals of various sizes (Fig. 1.6), while the green-black veins consist of m uscovite, epidote and metallic minerals. In the red limestone STE\_9 the color is attributed to the dif fusion of iron oxides in the micritic calcitic mass (Fig. 1.7). Green marbles on the other hand, like t he green cipolin marble STE\_4, owe their darkest green color to layers of mica, epidote, titanite an d metallic minerals (Fig. 1.8). Blackish marbles like dolomite P\_9 and limestone KM\_3, owe their color to the presence of bitoumens (Fig. 1.9, 1.10). Moreover, in sample KM\_3 concentrated micro crystals of iron-manganese oxides replace calcitic fossils (Fig. 1.10).







Figure 1 - Macroscopic image of the marble to the left and corresponding microscopic image of the marble to the right; calcitic marble NA\_1 (1), dolomitic marble AM\_3 (2), dolomitic-calcitic marble DM\_3 (3), calcitic marble DM\_1 (4), calcitic marble AM\_15 (5), calcitic marble TH\_2 (6), limestone STE\_9 (7), cipolin marble STE\_4 (8), dolomite P\_9 (9), limestone KM\_3 (10).

### 4. Geochemical analysis of the samples

Chemical analysis was performed in order to emphasize the differences or similarities in major and trace element trends between the marble groups studied. Major and trace element analyses results are given in table 1. Binary diagrams of the analyzed samples are shown in figure 3. In general, all marble groups display similar compositional distribution in terms of major oxide and trace element s. In this paper representative analyses of some samples are presented.

It is evident that all carbonate samples exhibit limited presence of major elements except Ca and Mg. The Mn, Ba and Sr elements and element ratios (i.e Mg/Ca, Mn/Sr, Ca/Sr) are much more useful than major elements alone for determining the geochemical affinities of rock (Koralaya *et al.*, 2016; Lazzarini *et al.*, 1980; Melezhik *et al.*, 2001; Melezhik *et al.*, 2008; Murra *et al.*, 2011; Bağcı *et al.*, 2010). In relatively pure marbles, the amounts of SiO2 and Al2O3 are very low and those of TiO2, Na2O and K2O are almost lacking. From the binary plots, it is shown a positive correlation between Na and K which is attributed to the limited presence of feldspars (Fig. 3g). All cipolin marbles are rich in Mn (STE\_4, STE\_5, STE\_6, STE\_7 and TH\_4), a fact that is attributed to layers of mica and epidote crystals between calcitic crystals. Limestones from Epirus region (H\_1-6), a limestone from Western Greece (DS-2) and limestones and marbles from Crete (K\_1-4, K\_7) are also Mn-rich. Cr3+ with ionic radius 0.62 Å substitutes Fe3+ (0.65 Å). Chromium is an element mainly connected to ultramafic rocks, especially in the grid of chromite or chromium silicate minerals (such as the chrome mica fuchsite). Samples from Peloponnesus (P\_4, P\_2, P\_7), Thessaly (TH\_3, TH\_4) and central Macedonia (KM\_3), appear to be rich in Cr3+. In the binary plots, there is a positive correlation between Ni and Cr observed

in the calcitic marbles, whereas in the dolomitic rocks the content of Ni does not seem to play any significant role in the extended presence of Cr. Dolomitic rocks contain minerals such as spinel in a standard amount; hence Ni and Cr cannot exceed a specific value. Moreover, there is a positive correlation between Cu and Zn in the calcitic marbles, whereas in the dolomitic rocks there is no statistically significant correlation between those two elements. As in the case of spinel, other similar heavy minerals that contain Cu and Zn have also a specific value range in dolomitic rocks.



Figure 2 - SEM microphotographs of marble samples. Minerals such as apatite (sample STE\_7a), fluorite (sample AM\_8b), sphalerite (sample AM\_8c) and epidote (sample TH\_2d) participate also in the mineral composition of the samples.

High Sb content occurs in limestone sample from western Greece (DS\_1: 6 ppm). Extremely high Ba co ntent has been found in samples from Sterea Ellada (calcitic-dolomitic marble STE\_1: 4800 ppm, dolom ite STE\_13: 6000 ppm) and in samples from Crete (dolomitic limestone K\_5: 4200 ppm, calcitic marble K\_1: 1600 ppm). As it is shown in the binary plots, there is a positive correlation between Ba and As, both in calcitic and dolomitic rocks, whereas there is no statistical correlation between Ba and Sr.

Strontium usually substitutes the Ca ions in dolomites due to its similar ion size. Thus it is expected that dolomitic rocks will present lower Sr contents than the calcitic rocks. Furthermore, the value range of Sr in calcitic rocks is higher than that in dolomitic, whereas marbles exhibit lower Sr value range compared to the non-metamorphic rocks (limestones, dolomites). Strontium is an indicator of diagenetic recrystallization in natural carbonate rocks. Its content decreases during dolomitisation and the subsequent replacement of calcite by dolomite (Shearman and Shirmohammadi, 1969; Garde, 1979). In addition, marbles that are in contact with schists tend to have slightly enriched Sr contents. The Sr/Ca ratio in seawater, the sedimentation environment, and biogenic factors affect strontium separation in biogenic carbonates and control the Sr/Ca ratio in marine carbonate rocks (Stoll and Schrag, 2001). Elements such as K, Na, Rb, and Sr are reported to be mobilized during sedimentation and

metamorphism. Since elements such as Th, Sc, and Y are least affected during sedimentation, they give more reliable information on the source properties (Chen *et al.*, 2002). The highest Rb contents appear in limestones (samples KM\_5: 92 ppm, K\_3: 92 ppm) and in calcitic marbles (sample K-5: 175 ppm). Rb<sup>+</sup> easily substitutes K<sup>+</sup> due to its chemical behavior and ion size (Rb<sup>+</sup>: 1.52 Å, K<sup>+</sup>:1.38 Å), so although Rb<sup>+</sup> does not form groundless minerals, it easily replaces K<sup>+</sup> in biotite and K-feldspars (Iliadou *et al.*, 2004).

wt.%	AM_13	STE_5	AM_17	AM_6	H_5	P_6	<b>STE_13</b>
SiO <sub>2</sub>	< 0.10	4.20	0.18	< 0.10	0.35	< 0.05	0.30
$Al_2O_3$	< 0.10	0.84	0.05	< 0.10	0.08	< 0.05	0.00
Fe <sub>2</sub> O <sub>3</sub>	< 0.10	0.61	0.05	< 0.10	0.20	< 0.05	0.31
CaO	54.80	52.15	32.80	30.90	53.80	54.00	35.10
MgO	< 0.05	0.80	19.80	21.10	0.30	0.65	17.80
MnO	< 0.05	0.05	< 0.05	< 0.05	< 0.05	< 0.05	0.00
TiO <sub>2</sub>	< 0.10	< 0.05	< 0.05	< 0.10	< 0.05	< 0.05	< 0.05
K <sub>2</sub> O	0.20	0.29	0.11	0.28	0.07	0.02	0.08
Na <sub>2</sub> O	1.06	0.32	0.65	1.09	0.35	1.30	0.39
LOI	42.81	41.10	45.15	46.24	43.50	42.44	46.80
ppm							
Cr	25.00	26.00	6.00	24.00	11.00	45.00	28.00
Mn	15.00	425.00	7.00	12.00	135.00	65.00	35.00
Ni	13.00	26.00	10.00	8.00	16.00	30.00	24.00
Cu	2.30	40.00	3.00	2.00	32.00	7.00	41.00
Zn	14.00	28.00	4.00	21.00	26.00	25.00	53.00
As	1.50	1.50	1.00	1.30	2.00	2.00	370.00
Sr	150.00	430.00	20.00	16.00	385.00	158.00	220.00
Ba	28.00	75.00	12.00	6.00	38.00	67.00	6000.00
AM_13 & STE_5 = calcitic marbles, AM_17 & AM_6 = dolomitic marbles ,dolo							

Table 1 - Chemical composition of representative samples.

#### 5. Conclusions

Samples of ornamental stones from different areas of Greece are examined. They are classified into three main categories; marbles (calcitic and dolomitic), limestones and dolomites, Particular emphasis is given on the correlation between mineralogical composition and aesthetic characteristics. The constituents that are responsible for these special features are the Fe-Mn oxides and silicate minerals such as quartz, muscovite, biotite, clinochlore, chlorite, epidote, clay minerals (illite, kaolinite), etc. Oxides are either diffused to the whole rock mass or are concentrated in veins, while the silicate minerals usually appear in strips or agglomerates. Moreover, the geochemical characteristics of the marble groups show that those which have undergone higher diagenetic processes (i.e. calcitic and dolomitic marbles) exhibit lower Sr contents in comparison to nonmetamorphic rocks (limestones and dolomites). Taking into consideration the fact that Sr substitutes only Ca, it is expected that all types of calcite-dominant rocks contain more Sr compared to the dolomitic ones. Positive correlations between Ba and As are observed in all studied groups, whereas there is no statistical correlation between Ba and Sr. Furthermore, it is concluded that dolomitic rocks show specific value ranges in their composition of the elements Ni, Cr, Cu and Zn, which cannot be correlated, while calcitic rocks show positive correlation and no limitation in their content. All these features affect marble appearance and contribute to their commercial final market price Considering also the fact that many of these ornamental stones have been used in architecture from ancient times as ornamental or building materials (statues, temples, emperors' residences, etc.), further investigation and research is required in order to enrich the existing database and archaeometric data which will be of immense importance for archeological and geoscience scientific fields.



Figure 3 - Binary diagrams of analyzed samples. CaO-Sr (a), Ba-Sr (b), Ba-As (c), Ni-Cr (d), Cu-Zn(e), K<sub>2</sub>O/Na<sub>2</sub>O-CaO/MgO (f), Na<sub>2</sub>O-K<sub>2</sub>O (g). The legend of the first diagram is valid for all diagrams.

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