Quality of soil and water in deltaic deposits of Louros and Arachthos rivers related to karstic rocks of the wider area

Papadopoulou K. National and Kapodistrian University of Athens, Faculty of Geology and Geoenvironment, Department of Geography & Climatology

Vriniotis D. Institute of Geological and Mineralogical Research

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QUALITY OF SOIL AND WATER IN DELTAIC DEPOSITS OF LOUROS AND ARACHTHOS RIVERS RELATED TO KARSTIC ROCKS OF THE WIDER AREA

Papadopoulou K., and Vriniotis D.

1 National and Kapodistrian University of Athens, Faculty of Geology and Geoenvironment, Department of Geography & Climatology, Panepistimioupoli, Zografou 15784, Athens, Greece
2 Institute of Geological and Mineralogical Research, Athens

Abstract

The present study focuses on the connection between the existing karstic formations and the quality of soil and water in the River Delta area. The hydrological basins of Louros and Arachthos Rivers (NW Greece) are partially developed on karstic rocks. So the soil is enriched with CaCO₃ and there is a natural supply of Ca²⁺, which originates mainly from the limestone and secondarily from gypsum formations. The value of the Cation Exchange Capacity (CEC) of soil is high and in the exchange sites Ca⁺⁺ ions dominate and Mg²⁺ secondarily.

Delta water of phreatic and confined aquifer contain high Ca⁺⁺ ion concentrations. The fluctuation of HCO₃⁻ values in confined aquifer water is within normal levels. Ions of SO₄²⁻ that occur in water are derived from the karstic process of gypsum and anhydrite formations. The values of Na⁺ and Cl⁻ content in water are higher in the phreatic aquifer than in ground water. The water in the area of the River Delta is classified as Carbonic water, despite its proximity to the sea.

Therefore, the quality of soil and water of the River Delta area is highly related to the karstic process of carbonate formations, mainly of calcareous and secondarily evaporate sediments.

Key words: Deltaic deposits, karstic rocks, water, soil, quality.

Περίληψη

Οι υδρογραφικές λεκάνες των ποταμών Λούρου και Αράχθου αναπτύσσονται κυρίως σε καρστικά πετρώματα από ασβεστόλιθους, δολομίτες και δευτερευόντως γύψου. Τα εδάφη στο δελτικό πεδίο των δύο ποταμών, όπως αναμενόταν, είναι επαρκώς εφοδιασμένα με CaCO₃ και υπάρχει φυσική τροφοδοσία σε Ca⁺⁺. Οι τιμές της Ικανότητας Ανταλλαγής (ΙΑΚ) στα εδάφη είναι υψηλές με κυριαρχία των εναλλακτικών ιόντων Ca⁺⁺ και δευτερευόντως Mg²⁺.

Στα νερά του Δέλτα τόσο του φρεατίου όσο και του υπόγειου υδροφόρου ορίζοντα η περιεκτικότητα σε ιόντα Ca⁺⁺ είναι υψηλή. Η διακύμανση του τιμών των ιόντων HCO₃⁻ στα νερά του υπόγειου υδροφόρου είναι σε φυσιολογικά επίπεδα. Τα υπάρχοντα SO₄²⁻ στα νερά προέρχονται από την καρστικοποίηση της γύψου και του ανυδρίτη. Οι τιμές των ιόντων Na⁺ και Cl⁻ είναι υψηλότερες στα νερά του φρεατίου
1. Introduction

The present study focuses on the quality of soil and water associated with the delta deposits of Louros and Arachthos Rivers and its relation to the presence of karst bedrock within the hydrological basins of the aforementioned rivers.

For this purpose, the CaCO$_3$ content in the soil of the river delta is examined. In addition, the Ca$^{++}$, HCO$_3^-$, SO$_4^{2-}$, Mg$^{++}$, Na$^+$ and Cl content in the phreatic and confined aquifer water of the River Delta area are also studied. The research was carried out under the 2$^{nd}$ Community Support Framework program were used (Vriniotis 2001).

Delta deposits of the Louros and Arachthos Rivers occur between the towns of Arta and Preveza, especially between the river beds of Louros on the west and Arachthos on the east. This plain covers an area of about 330 km$^2$, the highest altitude of which is 22 m. The plain is separated by the Valaora hill (246 m) on the east and by the Mavrovouni hill (+292 m) on the west.

The climate of the greater area is mild and changes to continental towards the highlands in the east. The lowest annual precipitation is less than 1,200 mm at the west coast of the Ionian Sea. Towards the delta area the precipitation is up to 1,600 mm, while towards the north-east, in the mountainous area reaches 2,600 mm (Boltsis 1986). The annual maximum and minimum temperature, as measured at the Aktio and Arta Meteorological Stations, is 37 °C and 41 °C, respectively, during the summer period, and −3.6 °C and −7.2 °C during the winter. In the area under study, the evaporation and humidity are high throughout the year. Precipitation, in combination with the recorded fluctuation in temperature, constitute the main factors that determine the physical – chemical process on the “rock – water – soil – flora – fauna” system.

2. Geological Setting

The geological formations that provide material to the alluvial deposits of the Louros-Arachthos River delta belong mainly to the Adriatic-Ionian Unit and secondarily part to the Pindos and Gavrovo-Tripoli Units (Map No 1). The hydrological basin of the Louros River is entirely developed on the geological formations of the Adriatic-Ionian Unit, while that of Arachthos River is extended and on formations of the Pindos and Gavrovo-Tripoli Units. Therefore, there is a differentiation on the composition of the weathered soil materials carried by the two rivers.

The geological formations of the Adriatic-Ionian Unit that are encountered within the hydrological basins of the rivers are as follows:

- Gypsum formations, which are mainly found within breccia of calcareous origin.
- Limestone formations of “Sinion” and “Pantokratora” of Upper Triassic-Liassic age. During the Jurassic period, the geotectonic zone was submerged and subdivided into three elongated parts, the axial, western and eastern sections. Schist and siliceous formations prevail in the axial area, while limestone is dominant at the edges. Later than Malmian, the “Vigla” limestone was deposited, while breccia with calcareous constituents was deposited till the Eocene period. Later on, flysch was deposited till Lower Miocene, whilst sandstone and marl were deposited during the Aquitanian-Burdigalian.

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The geological formations of the Gavrovo-Tripoli Unit are found on many locations within the hydrological basin of Arachthos River, towards E and NE. The alpine sediments are composed of upper Triassic dolomite, followed by Jurassic limestone, melanic Cretaceous limestone, and by cordonance limestone of Eocene epoch. The Eocene-Oligocene flysch covers the limestone series.

The geological formations of the Pindos Unit, with the overlying trusted ophiolites, supply with material the Arachthos River, especially the northern part of its tributaries. Dolomite and limestone of mid-Triassic age are the older formations of this Unit. The platy limestone and siliceous formations alternating with schist, and the schist-keratolithic complex formation of Lower Cretaceous age follow. The upward succession continues with marl, sandstone and breccia.
of Lower Cretaceous age and limestone of the Upper Cretaceous period are found. The aforementioned series of the Mesozoic period end with flysch until the beginning of Lower Oligocene age.

With respect to the post-alpine sediments, the following occur in the area:

- The “Pentalofos” molasse of Aquitanian-Burdigalian age.
- Neogene formations of Pliocene epoch, such as terrestrial conglomerate, clayey sand, white marl and marine clay with layers of gypsum. Gypsum formations are observed close to Palaeorofos village, as well as west of the Nea Filippiada settlement, north of the Amvrakikos Gulf. They are mainly found within a conglomerate zone, the direction of which is NNW-SSE. This formation is present between the Ziros Lake and the area located 4 km approximately north of the Nea Kerasounta settlement. These conglomerates are part of the Pleistocene sediments of Archagellos and Nea Filippiada settlements.
- Quaternary deposits are mainly found on the plain of the River Delta area. Their composition depends on the type of geological formations that supply with material the rivers, the distance of transfer, the water flow of rivers and other relevant factors.
- With respect to tectonism, the overthrust of the Units from east towards the west and the development of synclinal folds, alternating with anticlinal folds of NNW-SSE direction characterize the study area (Aubouin, 1959). Faults mainly follow the NNW-SSE direction, while there are several faults - usually more recent - that follow the E-W direction. The faults are created by the “tectonism of breaks” that followed the “tectonism of thrusting – up thrusting”. Basic element of the tectonism, acting on the greater area, is the presence of gypsum, which constituted the sliding plane and developed phenomena of diapirism.

3. Geomorphology

The hydrological basin of Arachthos River is developed NNE of the area under study and covers an area of 1,850 km$^2$. The main river bed of Arachthos River is 105 km long. The hydrological basin of Louros River is developed north of the area under study and covers an area of 685 km$^2$. The length of the main Louros river bed is 73 km. The main river beds of Arachthos and Louros Rivers follow a NNW – SSE direction, similar to the direction of the fold axes of the geological formations prevailing in the greater area and of the Hellenic geotectonic zones (Mertzianis 1992).

The hydrological basin of Arachthos River is made up of geological formations prone to erosion (flysch, marl, sandstone), which produce sediments, as well as karst formations (limestone), constituting 18.5% of surface exposures.

The hydrological basin of Louros River is made of limestone (approximately 70%).

Therefore, there exists a quantitative and qualitative differentiation of sediments carried by the two rivers. The percentage of alluvium supplied by the Arachthos River is greater than the one supplied by Louros River. The Arachthos River delta is adequately developed. Arachthos River has greatly contributed to the development of the plain of Arta (Hellenic Marine Research Centre 1989).

The Arachthos River provides coarse grain material, mainly quartz, and also large quantities of fine-grained silt. They originate from the flysch and the ultramafic geological formations occurring in the upstream hydrological basin of Arachthos River.

The Louros River supplies small quantities of fine-grained materials, as well as angular or rounded gravel of calcareous origin.

The calcareous deposits of Louros River found in the western part of the study area consist of clayey silt. On the contrary, the deposits of Arachthos River found in the eastern-central part are mostly made of sandy silts (Vriniotis 2001).
4. Methodology

4.1. Geochemistry of Soil

Soil sampling has been carried out on cultivated land that covers an area of 330 km$^2$. Eighty-seven soil samples were collected from the river delta area. The sample density was one soil sample per 4 km$^2$. Sampling locations were pre-determined. Two samples of 1.5-2.0 kg weight approximately were collected from each sampling location. The soil samples were collected down to a depth of 30 cm. Samples were left to dry and then sieved through the <10 mesh. Several analyses were performed on these soil samples, including the determination of CaCO$_3$ by the Bernard calcium-meter and Cation Exchange Capacity (CEC) by atomic absorption.

The results of the analyses were statistically processed by use of ARC – INFO and SURFER software packages. In detail, several statistical parameters were calculated, histograms plotted, and grain-size analysis was carried out.

4.2. Geochemistry of Water

Seventy-four water samples were collected, forty-five from the confined aquifer of the area, and twenty-nine from the phreatic.

The confined aquifer water samples were collected from operating boreholes between 35 and 75 m depth and are distributed evenly over the delta area as far as possible close to the soil sampling locations.

Water samples from the phreatic aquifer were collected from shallow water wells, most of which were not in use. The sampling depth was approximately 9 m below surface.

The volume of each water sample was 1,000 cm$^3$, and the analyses performed included Ca$^{++}$, HCO$_3^-$, SO$_4^{--}$, Mg$^{++}$, Cl$^-$ and Na$^+$ ions. The ions of Ca$^{++}$, Mg$^{++}$, HCO$_3^-$ and Cl$^-$ were determined volumetrically by titration, SO$_4^{--}$ ion by the spectrometer Hach 4000, and Na$^+$ by atomic absorption PE 2100.
In the area under study, the CEC of all samples is >12.4 meq/100 g of soil, with an average of 30.8 meq/100 g of soil (Fig. 1). These high values are mainly attributed to the type of clay minerals and organic matter content. In the exchange sites Ca$^{2+}$ ions dominate and secondarily Mg$^{2+}$, resulting in the development of favorable conditions for soil. The source of this Ca$^{2+}$ content in soil is primarily the limestone formations and to a minor degree the gypsum formations.

5.2. Water

5.2.1. Ca$^{2+}$ Cation Content in Water

Most of the Ca$^{2+}$ cations are derived primarily from the hydrolysis of CaCO$_3$ and partially from gypsum formations. As it has already been mentioned, the bedrock and the soil in the study area are rich in CaCO$_3$. Moreover, there are in the area occurrences and deposits of gypsum.

The Ca$^{2+}$ content in the water of the phreatic aquifer is up to 13.44 meq/l (268 mg/l), with an average value of 5.41 meq/l, whilst in the water of the confined aquifer is up to 6 meq/l (120 mg/l), with an average of 4.18 meq/l (Map No 2). So, water drawn from boreholes is suitable for drinking purposes, whereas that of water from wells is classified as good to moderate, based on the Waterlot diagram.

The increase CO$_2$ content in the phreatic aquifer, which originates from organic matter or from precipitation, leads to a higher Ca$^{2+}$ content in the phreatic aquifer water compared to the confined aquifered water. Moreover, in areas supplied by alluvium from Louros River, close to calcareous formations, the Ca$^{2+}$ content is high in phreatic aquifer too.

5.2.2. Bicarbonate (HCO$_3^-$) Content in Water

The bicarbonate ions are mainly derived from the hydrolysis of CaCO$_3$ in the Delta area. The bicarbonate content in the water of the confined aquifer reaches a maximum value of 8.07 meq/l (496 mg/l), with an average of 4.54 meq/l. This variation is considered to be within the normal range (Table 1). The HCO$_3^-$ content of the water in the phreatic aquifer reaches up to 23.44 meq/l (1,453 mg/l), with an average of 6.70 meq/l. The values above 9 meq/l are to be considered anomalous. This is attributed to the decomposition of organic matter in soil, which increase the CO$_2$ content.
### Table 1 - Consistency (%) of elements in Water (Vriniotis 2001)

<table>
<thead>
<tr>
<th>HCO₃ (meq/l)</th>
<th>%</th>
<th>Phreatic Aquifer</th>
<th>Ground Water Table</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>2.64-3.19</td>
<td>13.8</td>
<td>1.36</td>
</tr>
<tr>
<td>2</td>
<td>3.20-4.21</td>
<td>31.0</td>
<td>number of samples = 28</td>
</tr>
<tr>
<td>3</td>
<td>4.22-7.34</td>
<td>17.2</td>
<td>Mean = 6.70</td>
</tr>
<tr>
<td>4</td>
<td>7.35-9.69</td>
<td>17.2</td>
<td>Background = 5.38</td>
</tr>
<tr>
<td>5</td>
<td>9.70-16.90</td>
<td>17.2</td>
<td>Standard deviation = 4.45</td>
</tr>
<tr>
<td>6</td>
<td>16.91-23.44</td>
<td>3.4</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Mg²⁺ (meq/l)</th>
<th>%</th>
<th>Phreatic Aquifer</th>
<th>Ground Water Table</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>0.08 – 1.52</td>
<td>21.4</td>
<td>9.1</td>
</tr>
<tr>
<td>2</td>
<td>1.53 – 2.07</td>
<td>17.9</td>
<td>number of samples = 45</td>
</tr>
<tr>
<td>3</td>
<td>2.08 – 3.75</td>
<td>21.4</td>
<td>Mean = 3.40</td>
</tr>
<tr>
<td>4</td>
<td>3.76 – 6.77</td>
<td>21.4</td>
<td>Background = 2.50</td>
</tr>
<tr>
<td>5</td>
<td>6.78 – 9.09</td>
<td>14.3</td>
<td>Standard deviation = 2.65</td>
</tr>
<tr>
<td>6</td>
<td>9.10 – 9.12</td>
<td>3.6</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>CI (meq/l)</th>
<th>%</th>
<th>Phreatic Aquifer</th>
<th>Ground Water Table</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>0.35-0.56</td>
<td>18.5</td>
<td></td>
</tr>
<tr>
<td>2</td>
<td>0.57-1.20</td>
<td>14.8</td>
<td>number of samples = 28</td>
</tr>
<tr>
<td>3</td>
<td>1.21-3.71</td>
<td>37.0</td>
<td>Mean = 4.54</td>
</tr>
<tr>
<td>4</td>
<td>3.72-7.89</td>
<td>14.8</td>
<td>Background = 2.00</td>
</tr>
<tr>
<td>5</td>
<td>7.90-16.76</td>
<td>11.1</td>
<td>Standard deviation = 4.75</td>
</tr>
<tr>
<td>6</td>
<td>16.77-17.80</td>
<td>3.7</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Na⁺ (meq/l)</th>
<th>%</th>
<th>Phreatic Aquifer</th>
<th>Ground Water Table</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>0.26-0.47</td>
<td>11.5</td>
<td></td>
</tr>
<tr>
<td>2</td>
<td>0.48-1.38</td>
<td>11.5</td>
<td>number of samples = 28</td>
</tr>
<tr>
<td>3</td>
<td>1.39-4.04</td>
<td>50.0</td>
<td>Mean = 4.22</td>
</tr>
<tr>
<td>4</td>
<td>4.05-8.23</td>
<td>23.1</td>
<td>Background = 2.61</td>
</tr>
<tr>
<td>5</td>
<td>8.24-16.78</td>
<td>7.4</td>
<td>Standard deviation = 6.35</td>
</tr>
<tr>
<td>6</td>
<td>16.79-34.20</td>
<td>3.6</td>
<td></td>
</tr>
</tbody>
</table>

### 5.2.3. Sulphate Ion (SO₄²⁻) Content in Water

The sulphate ion content has values up to 350 mg/l, with an average of 2 meq/l (100 mg/l). There were only three sampling locations of the phreatic aquifer where the sulphate ion level in water was greater than 250 mg/l. The highest values were measured in water collected from the area NE of Rodia and Loutrotopos, as well as from the area of the Kerasounta settlement. This constituent originates from the karstified gypsum formations (CaSO₄·2H₂O) and anhydrite (CaSO₄) of the Neogene evaporites occurring in the area.

### 5.2.4. Mg⁺⁺, Na⁺ and Cl⁻ Content in Water

The results of the analyses are presented in Table 1. Mainly the Na⁺ and Cl⁻ ion contents and secondarily Mg⁺⁺ are higher in the phreatic aquifer water in comparison to that of confined aquifer. This is partially attributed to the transportation of Na⁺: Mg⁺⁺ and Cl⁻ ions by sea spray with the...
airmasses. So karst water of the confined aquifer is not affected by the sea and is of better quality, compared to that of the phreatic aquifer.

In general, water can be classified into the following two types:

1. Water containing carbon
   \[ \text{HCO}_3^- > \text{SO}_4^{2-} > \text{Cl}^- \]
   \[ \text{Ca}^{++} > \text{Mg}^{++} > \text{Na}^+ \]

2. Water containing chlorides
   \[ \text{Cl}^- > \text{SO}_4^{2-} > \text{HCO}_3^- \]
   \[ \text{Na}^+ > \text{Mg}^{++} > \text{Ca}^{++} \]

According to the above water types (i) and (ii), the geochemical category in which a sample of water can be classified, depends upon the decreasing content of ions and cations, especially on the type of the dominant ion (Kallergis 1968).

Based on the aforementioned results, although the delta water is near the sea, it mainly belongs to the class of Carbonate water. Therefore, water quality is strongly associated with the karstic process of the calcareous formations, prevailing in the study area.

6. Conclusions

Based on the above results, the following are concluded:

- Soil in the area of Louros and Arachthos River Delta is sufficiently supplied with \( \text{CaCO}_3 \). This constituent originates from the calcareous formations occurring in the greater area of the hydrological basins, mainly that of Louros.
- Soil in the area of Louros and Arachthos River Delta is naturally supplied with \( \text{Ca}^{++} \), which originates from the existing karstified geological formations, mainly from the limestone and secondarily from gypsum.
- The value of the Cation Exchange Capacity of soil is rather high, and is dominated by the high values of exchangeable \( \text{Ca}^{++} \) ions.
- In the delta water, both in the phreatic and confined aquifer, a high \( \text{Ca}^{++} \) ion content is observed. The \( \text{Ca}^{++} \) ion content is higher in the water of the phreatic aquifer of the area, especially in the adjacent calcareous formations, where alluvial materials are deposited by Louros River.
- The variation of \( \text{HCO}_3^- \) content in water of the confined aquifer is within normal levels. In general, the \( \text{HCO}_3^- \) ions are mainly formed by the hydrolysis of \( \text{CaCO}_3 \), which occurs in large amounts due to the presence of calcareous formations. The values of the two higher classes in the water of the phreatic aquifer are rather high, which is attributed to the decomposition of organic matter in soil.
- The \( \text{SO}_4^{2-} \) ion content in the delta water is derived from the karstified formations of Neogene evaporites occurring in the greater area.
- The values of \( \text{Na}^+ \) and \( \text{Cl}^- \) content in water are higher in the phreatic aquifer, whereas the confined aquifer water is not affected by the sea and is of better quality.
- The delta water should be classified as Carbonate water and the potability of borehole water (confined aquifer), with respect to the \( \text{Ca}^{++} \), \( \text{Mg}^{++} \), \( \text{Na}^+ \), \( \text{Cl}^- \), \( \text{SO}_4^{2-} \) and \( \text{HCO}_3^- \) ion content is good whilst that of the phreatic aquifer is classified as good to moderate.
- It is concluded that there is a close association between the quality of soil and water in the deltaic deposits of Louros and Arachthos Rivers with the occurrence of carbonate formations.
7. References


