

## Mediterranean Marine Science

Vol 14, No 2 (2013)

Vol 14, No 2 (2013)



### Volatile compounds of some popular Mediterranean seafood species

*I. GIOGIOS, N. KALOGEROPOULOS, K. GRIGORAKIS*

doi: [10.12681/mms.342](https://doi.org/10.12681/mms.342)

#### To cite this article:

GIOGIOS, I., KALOGEROPOULOS, N., & GRIGORAKIS, K. (2013). Volatile compounds of some popular Mediterranean seafood species. *Mediterranean Marine Science*, 14(2), 343–352. <https://doi.org/10.12681/mms.342>

## Volatile compounds of some popular Mediterranean seafood species

I. GIOGIOS<sup>1,2</sup>, N. KALOGEROPOULOS<sup>1</sup> and K. GRIGORAKIS<sup>2</sup>

<sup>1</sup>Department of Nutrition & Dietetics, Harokopio University, 17671, Athens, Greece

<sup>2</sup>Hellenic Centre for Marine Research, Agios Kosmas Hellinikon, 16777, Athens, Greece

Corresponding author: [nickal@hua.gr](mailto:nickal@hua.gr)

Handling Editor: Argyro Zenetos

Received: 6 February 2012; Accepted: 13 April 2013; Published on line: 11 June 2013

### Abstract

The volatile compounds of highly commercialised fresh Mediterranean seafood species, including seven fish (sand-smelt *Atherina boyeri*, picarel *Spicara smaris*, hake *Merluccius merluccius*, pilchard *Sardina pilchardus*, bogue *Boobps boops*, anchovy *Engraulis encrasicolus* and striped-mullet *Mullus barbatus*), squid (*Loligo vulgaris*), shrimp (*Parapenaeus longirostris*) and mussel (*Mytilus galloprovincialis*), were evaluated by simultaneous steam distillation-extraction and subsequent GC-MS analysis. A total of 298 volatile compounds were detected. The mussels contained the highest total concentration of volatile compounds, while pilchard among fish species contained the highest number and concentrations of volatile compounds. Individual patterns of volatile compounds have been distinguished. The fish species when compared to the shellfish species studied, contained 6 to 30 times more 1-penten-3-ol, higher quantities of 2-ethylfuran, and 2,3-pentanedione, which was absent from the shellfish species. Pilchard is characterized by a high concentration of alcohols, shrimps by the high presence of amines and S-compounds, while mussels by high amounts of aldehydes, furans, and N-containing compounds (pyridine, pyrazines and pyrroles). The fatty acid-originating carbonyl compounds in fish seem to be related to the species' fat content.

**Keywords:** Volatile compounds, flavour, Mediterranean fish, shrimp, squid, mussel.

### Introduction

Volatile compounds largely define the flavour and subsequently the general organoleptic quality of seafood (Kawai, 1996; Ólafsdóttir & Jónsdóttir, 2010). They are those characterizing either species-related delicate fresh fish odours or spoilage related odours (Josephson *et al.*, 1984; Ólafsdóttir & Jónsdóttir, 2010).

The volatile compounds of various fresh fish (indicatively, Kawai, 1996; Prell & Sawyer, 1988; Morita *et al.*, 2003) and other raw seafood (Chung, 1999; Piveteau *et al.*, 2000; Pennarun *et al.*, 2002; Zhang *et al.*, 2010) have been studied. Sporadic data occur in the literature regarding the volatiles of Mediterranean fish species, referring mainly to sardine (Triqui & Bouchriti, 2003; Prost *et al.*, 2004), gilthead sea bream (Grigorakis *et al.*, 2003; Alasalvar *et al.*, 2005) and European sea bass (Leduc *et al.*, 2012). With respect to Mediterranean shellfish, some literature is available for mussel volatiles (Rasmussen *et al.*, 1993; Le Guen *et al.*, 2000a, b; Fuentes *et al.*, 2009).

Fish and other seafood share many similar volatile compounds, contributing to similarly general aroma notes, but they also possess species-specific aromas (Morita *et al.*, 2003). The aim of this study was to compare the volatile compounds of some very common, highly marketed Mediterranean seafood species. This

would definitely add to the knowledge regarding the flavour of seafood produced and consumed within this region. However, the importance of such knowledge is more general since most of these species are consumed world-wide. The choice of the studied species was based on their market popularity, i.e. the selected species are all widely consumed in the Mediterranean area, and on the fact that each one of them is traditionally cooked using several diverse methods. Therefore, this study aimed to provide data that would serve as a basis for further study on the effects of various cooking methods on seafood volatile compounds. Besides, excluding sardine and mussel, no data are available for the volatile compounds of the rest of the seafood species studied herein

### Material and Methods

#### *Specimens*

The fish and shellfish species used for this study were purchased in January – February 2011 from the local fish market in Athens (Greece), while mussels were obtained from an aquaculture unit. The quantity of each sample was 2-4 kg, comprising individuals of commercial sizes, freshly harvested and stored on ice according to the hygiene regulations and requirements.

Mussels were kept alive. All seafood was of the same

freshness stage, characterized as Extra according to the EU freshness grading (Howgate *et al.*, 1992), thus implying that they were all received within a few hours post-harvest. Somatometric characteristics and origin of the studied species appear in Table 1. Adequate quantities were obtained, and upon arrival at the laboratory, samples were homogenized, wrapped in aluminium foil, which was subsequently sealed in plastic bags, in order to avoid transfers of volatile compounds from the air or from contact with materials, and immediately stored until analysis, at -80°C, in order to prevent further post-mortem formation of volatile compounds.

### **Extraction of volatile compounds**

Volatile compounds were extracted from seafood tissues by simultaneous steam distillation - extraction (SDE). For this purpose, tissue samples of approx. 30 g, were homogenized with 60 mL 30% w/v NaCl solution prepared with double-distilled water, containing 100 ppm BHT as antioxidant. The homogenate was transferred to a Likens - Nickerson apparatus and was distilled-extracted with 25 mL CH<sub>2</sub>Cl<sub>2</sub> (HPLC grade, Waters, Milford, MA). The solvent temperature was kept at 50-55°C by a water-bath. Gradual and homogenous temperature increase and gentle boiling was applied by the addition of a few boiling chips, constant magnetic stirring and heating by a mantel. Cooling of the extraction system was achieved with ethylene glycol kept at -5 °C. The extraction lasted 3 hours; thereafter, heating was interrupted and the system was left for 30 min to equilibrate. The organic phase was collected, dried by anhydrous Na<sub>2</sub>SO<sub>4</sub> and subsequently concentrated first to 1-3 mL by a Kuderna-Danish apparatus and further to a volume of 0.5 mL by gentle nitrogen flow. The extracts were sealed in GC vials and stored at -20 °C until analysis. Two extractions per sample were conducted.

### **Volatile compounds analysis**

An Agilent 7890 GC, equipped with an Agilent 5975C mass spectrometer (Agilent Technologies, Santa Clara, CA, USA) was employed for the analysis of volatile compounds, which were separated on an Agilent HP-5MS capillary column (30 m × 0.25 mm, coated with a 0.25 µm film thickness of 5% phenyl-95% methylsiloxane). For the analysis, 2 µL of each extract was injected into the GC/MS at splitless mode. Helium was used as a carrier gas at a flow rate of 1.8 mL/min. The injector and transfer line temperatures were 200 and 300 °C, respectively. The oven temperature program was: initially 35 °C, increased to 95 °C at 2 °C/min, then to 195 °C at 8 °C/min and finally to 240 °C at 12 °C/min, where it was held for 4 min.

The mass spectrometer was set at 70 eV in EI mode, and the detector operated with a mass range of 30-300 amu, with the repetition rate set at 2.73 scans/sec. Peaks identification was made by employing NIST 05 Mass

Spectral Library (Rev. D.05.01) (Agilent Technologies, USA) together with the peak deconvolution software AMDIS (Version 2.64), as well as by referring to the mass spectra of standards, that were used for recoveries estimation. Recoveries were calculated by applying the whole procedure in a mixture of standards comprised of (R)-(+)-limonene 97% (Sigma), nonanal 95% (Aldrich), 1-penten-3-ol 99% (Aldrich), 2,4 heptadienal (technical grade) 90% (Aldrich), pentadecane ≥99% (Aldrich), 1-heptanol (puriss. p.a., standard for GC, ≥99% (GC) (Fluka), 4-heptanone purum, ≥96% (GC) (Fluka), benzene anhydrous, 99.8% (Sigma- Aldrich), and toluene anhydrous, 99.8% (Sigma- Aldrich). Reference curves were constructed for each of the aforementioned standards (Table 2). Recoveries were calculated after standard addition of five increasing concentrations (0.1, 0.2, 0.5, 1 and 2 µg /Kg) of the reference compounds in 1 g samples of white fish muscle (gilthead sea bream, *Sparus aurata*), by following the whole SDE procedure and by subtracting, from the areas found, the respective areas obtained from the single fish-muscle sample; the recoveries obtained are shown in Table 2. Each volatile compound was subsequently quantified by comparing its peak area to the standard curve of a chemically similar reference compound (Table 2), and concentrations were corrected accordingly, taking into account the recoveries of the respective quantifying reference compounds. A 200 µg/L solution of isooctane, >99.0% (Agilent), was added to every sample just before SDE extraction to serve as internal standard for the quantification of all those volatile compounds not mentioned in the quantification list of Table 2. In this case, quantification took place by comparing the compounds peak area to that of the internal standard, essentially as described by Alasalvar *et al.* (2005).

Blanks were always run between samples in order to ensure that no volatile compounds occurred besides those characterizing the studied seafood.

### **Statistics**

Cluster analysis was conducted to evaluate the proximity of the studied species as regards the profiles of their volatile compounds, using the Squared Euclidian Distance. The significance of the individual volatile compounds for grouping the samples was evaluated by factor analysis, employing the principal components extraction method. Because of the high number of identified compounds, only those present in at least half of the studied species have been evaluated. All the statistical analyses were conducted by means of the statistical program SPSS 13.0.

### **Results and Discussion**

Among a total of 298 volatile compounds, which were detected in the seafood samples, the 75 most charac-

**Table 1.** Common and scientific names, somatometric characteristics and origin of the studied species.

Common name	Scientific name	Weight (g) <sup>a</sup>	Total length (cm) <sup>a</sup>	Origin
<i>Fish</i>				
Big scale sand smelt	<i>Atherina boyeri</i>	4.2 ± 1.3	7.5 ± 0.9	Leros island (SE Aegean)
Picarel	<i>Spicara smaris</i>	14.1 ± 2.3	9.5 ± 0.5	Evoikos Gulf (W Aegean)
European hake	<i>Merluccius merluccius</i>	43.6 ± 11.4	16.6 ± 1.5	Lesvos island (NE Aegean)
European pilchard (or sardine)	<i>Sardina pilchardus</i>	18.3 ± 5.1	10.6 ± 0.8	Kavala (N Aegean)
Bogue	<i>Boops boops</i>	128.6 ± 8.0	17.5 ± 0.5	Chios island (NE Aegean)
European anchovy	<i>Engraulis encrasicolus</i>	12.5 ± 1.8	10.4 ± 0.6	Evoikos Gulf (W Aegean)
Striped mullet	<i>Mullus barbatus</i>	17.9 ± 5.1	9.2 ± 0.7	Kavala (N Aegean)
<i>Other seafood</i>				
Squid	<i>Loligo vulgaris</i>	68.1 ± 15.1	18.4 ± 1.9	Chios island (NE Aegean)
Shrimp	<i>Parapenaeus longirostris</i>	7.9 ± 2.4	12.1 ± 1.2 <sup>b</sup>	Saronikos Gulf
Mussel	<i>Mytilus galloprovincialis</i>	22.2 ± 6.2 <sup>c</sup>	6.2 ± 0.5 <sup>c</sup>	Saronikos Gulf (aquaculture)

<sup>a</sup> Average ± standard deviation, obtained from 20-40 individuals; <sup>b</sup> Includes cephalothorax; <sup>c</sup> Includes shell.

teristic and most abundant (found at concentrations >0.1 µg/kg) are shown in Table 3. Based on the fact that all species were extra fresh it is presumed that all volatiles found in the studied seafood are those characterizing the species, or some of them can alternatively be of environmental origin, especially in the case of filter feeders such as mussels (Fuentes *et al.*, 2009). In the case of seawater

fish, it has been shown that feeding history has some effects on volatile compound profiles when wild and cultured fish of the same species were compared (Grigorakis *et al.*, 2003; Alasalvar *et al.*, 2005). On the other hand, contradicting findings occur in studies comparing diets in farmed fish, showing minor or no effects in some cases (Grigorakis *et al.*, 2009; Silva *et al.*, 2012) and some ef-

**Table 2.** Standard curves (slope *a* and *R*<sup>2</sup>) and recoveries (%) for the authentic standards used for quantification of the volatile compounds.

Authentic standard	<i>a</i>	<i>R</i> <sup>2</sup>	Recovery %	Volatile compounds quantified by means of each authentic standard
2-Butanone	23074	0.9938	98.27	Ketones with retention times from 2.9 to 7.3 min
Benzene	31939	0.9799	82.15	Benzene, alkyl-benzenes, p-xylene, o-xylene, ethers, amines, furans
1-Penten-3-ol	25391	0.9895	97.66	Non-aliphatic alcohols
Toluene	30421	0.9806	77.07	Toluene, alkyl-toluenes, pyrazines, pyridines, pyrroles, pyrrolidines
4-Heptanone	28916	0.9850	99.84	Ketones with retention times from 7.31 to 56.3 min, esters
1-Heptanol	10643	0.9838	99.90	Aliphatic alcohols
D-Limonene	18866	0.9834	99.90	Terpenes, alicyclic hydrocarbons
Nonanal	9167.7	0.9635	99.90	Aldehydes
Octadecane	5970.3	0.9344	60.27	Alkanes, alkenes, alkadienes

**Table 3.** Main volatile compounds\* ( $\mu\text{g}/\text{kg}$  fresh weight) characterizing the studied Mediterranean seafood species. The retention times (Rt) and characteristic ions for each compound are also provided (nd stands for not detected; tr for traces).

Compound	Rt (min)	Characteristic ions (m/z)	<i>A. boyeri</i>	<i>S. smaris</i>	<i>M. merluccius</i>	<i>S. pilchardus</i>	<i>B. boops</i>	<i>E. encrasicolus</i>	<i>M. barbatus</i>	<i>L. vulgaris</i>	<i>P. longirostris</i>	<i>M. galatoprovincialis</i>
Tetrahydrofuran	3.042	42,72,71	0.06	0.05	0.04	0.17	0.04	nd	0.23	0.13	0.06	0.82
<i>trans</i> -2-Butenal	3.279	70,39,69	nd	nd	Nd	0.22	nd	nd	0.02	nd	nd	0.29
3-Methylbutanal	3.298	44,41,43	3.01	0.39	0.22	1.43	0.27	1.30	0.96	1.14	0.45	3.94
2-Methylbutanal	3.395	41,57,58	1.38	nd	0.18	0.72	nd	0.56	0.56	nd	nd	2.49
N,N,N',N'-Tetramethyl-methanedi-amine	3.432	58,42,44	nd	nd	nd	nd	nd	nd	nd	nd	0.35	nd
1-Penten-3ol	3.590	57,41,39	1.33	0.88	0.60	2.63	0.60	0.77	0.95	0.06	0.13	0.12
1-Methoxy-2-propanol	3.612	45,47,43	nd	0.49	nd	4.48	nd	nd	nd	0.82	nd	nd
3-Penten-2-ol	3.633	71,43,41	0.03	0.17	0.02	nd	nd	nd	nd	0.01	nd	nd
2-Pentanone	3.694	43,45,48	nd	nd	nd	0.14	0.03	0.03	nd	nd	0.02	nd
2,3-Pentanedione	3.767	43,57,100	0.21	0.27	0.13	1.03	0.21	0.46	0.23	nd	nd	nd
Heptane	3.767	43,71,56	nd	nd	nd	nd	nd	nd	nd	0.23	nd	1.91
Pentanal	3.785	44,58,45	0.45	0.30	0.28	0.84	0.26	0.32	0.29	0.06	nd	3.78
2-Ethyl-furan	3.822	81,96,53	0.72	0.41	0.28	2.31	0.42	0.57	0.44	0.02	0.02	0.15
3-Hydroxy-2-butanone	3.987	45,43,88	1.19	2.18	1.42	1.17	0.50	3.29	1.63	1.92	0.81	0.88
3-Methyl-1-butanol	4.432	57,70,42	0.11	0.02	nd	nd	nd	nd	0.08	0.01	nd	nd
Pyridine	4.624	79,52,51	nd	0.03	0.01	0.01	0.03	nd	0.02	0.04	0.07	1.54
2-Methoxy-furan	4.694	98,83,69	nd	nd	nd	nd	nd	nd	nd	nd	nd	0.20
<i>cis</i> -2-Pentenal	4.846	55,83,84	0.29	0.26	0.20	0.91	0.22	0.23	0.28	nd	nd	nd
Pyrrrole	4.919	67,41,39	nd	nd	nd	nd	nd	nd	0.04	nd	0.03	0.13
Toluene	5.035	91,92,65	0.33	0.15	0.23	11.12	0.01	2.06	6.90	3.49	0.29	22.24
1-Pentanol	5.125	42,55,41	0.03	0.01	nd	nd	0.01	nd	0.01	nd	nd	0.21
<i>trans</i> -2-Penten-1-ol	5.188	57,41,44	0.07	0.07	nd	0.24	0.04	tr	0.02	nd	nd	nd
3-Methyl-2-butenal	5.596	84,55,41	nd	nd	nd	nd	nd	nd	nd	nd	nd	0.40
Octane	5.901	43,85,57	0.14	0.22	0.23	0.21	0.12	0.09	0.15	0.08	0.04	2.14
Hexanal	5.962	44,56,41	0.25	0.38	0.19	1.03	0.40	0.12	0.33	0.02	nd	0.49
3-Pentanol	6.029	59,107,45	nd	0.03	0.02	0.10	0.02	0.04	0.05	0.02	nd	0.07
2-Hydroxy-3-pentanone	6.248	45,57,59	0.11	0.01	0.03	0.05	0.01	0.05	0.03	nd	nd	0.04
Unknown	6.474	58,42,61	nd	nd	nd	nd	nd	nd	nd	0.56	1.70	nd
2-4-Dimethyl-heptane	6.596	43,85,57	0.15	0.25	0.12	0.50	0.21	0.25	nd	nd	nd	0.82
Furfural	7.059	96,95,39	0.03	0.01	0.01	0.02	0.02	0.05	0.01	0.04	tr	0.36
3-Methoxy-1-butanol	7.096	59,43,89	nd	nd	nd	0.42	nd	nd	nd	nd	nd	nd
2-4-Dimethyl-1-heptene	7.285	43,70,55	0.05	0.08	0.04	0.09	0.05	0.10	0.02	0.05	0.02	0.27
2-Methyl-1H-pyrrole	7.352	80,81,53	nd	nd	nd	nd	nd	nd	nd	nd	nd	0.18
2-Hexenal	7.803	41,69,55	0.12	0.11	0.05	0.50	0.08	0.11	0.08	nd	nd	0.09
2-Furanmethanol (furfuryl alcohol)	7.980	98,97,41	0.14	nd	nd	nd	0.01	nd	nd	nd	nd	1.40
4-Methyl-octane	8.163	43,85,71	0.15	0.19	0.11	0.35	0.17	0.23	0.02	0.10	0.04	0.62
<i>p</i> -Xylene	8.345	91,106,105	0.04	nd	nd	0.02	0.02	0.06	nd	0.02	nd	0.26
1-Methoxy-2-propyl-acetate	8.716	43,45,72	nd	nd	nd	0.13	nd	nd	nd	0.06	nd	nd
Styrene	9.345	104,103,78	nd	0.01	0.06	0.08	nd	0.03	0.27	0.16	0.01	0.06
4-Heptanone	9.449	43,58,71	0.02	0.02	0.02	0.06	0.01	0.01	0.03	nd	0.02	0.14
<i>trans</i> -4-Heptenal	9.796	41,55,68	0.17	0.12	0.08	0.45	0.05	0.08	0.05	0.02	nd	0.25
1-(2-Furanyl)-ethanone	10.382	95,100	tr	tr	tr	nd	nd	0.01	nd	nd	nd	0.15
2-3-Dimethyl-pyrazine	10.746	108,67,40	nd	nd	nd	nd	nd	nd	nd	nd	0.03	0.02
Unknown	10.894	57,42,86	nd	nd	nd	nd	nd	nd	nd	nd	0.212	nd
Unknown	11.827	95,126,96	0.09	nd	nd	0.03	0.01	nd	nd	nd	nd	0.13
5-Methyl-2-furancarboxaldehyde	12.296	110,109,53	0.26	0.01	nd	0.03	0.07	nd	nd	nd	nd	0.48
Benzaldehyde	12.790	106,105,77	0.21	0.11	0.15	0.26	0.14	0.15	0.12	0.15	0.12	0.96

(continued)

(continued)

Compound	Rt (min)	Characteristic ions (m/z)	<i>A. boyeri</i>	<i>S. smaris</i>	<i>M. merluccius</i>	<i>S. pilchardus</i>	<i>B. boops</i>	<i>E. encrasicolus</i>	<i>M. barbatus</i>	<i>L. vulgaris</i>	<i>P. longirostris</i>	<i>M. galloprovincialis</i>
2-thiophenecarboxaldehyde	14.564	111,112,83	0.15	0.04	0.03	0.07	0.02	0.04	0.10	0.04	nd	0.09
2-Pentyl furan	14.771	81,82,138	0.03	0.02	0.03	0.09	0.04	0.02	0.04	nd	nd	0.03
2-Octanone	14.850	43,58,59	nd	nd	nd	nd	nd	nd	nd	nd	nd	0.04
2-Ethyl-6-methyl-pyrazine,	15.210	121,122,39	nd	nd	nd	nd	nd	nd	nd	nd	nd	0.01
<i>trans</i> -2-(2-Pentenyl) furan	15.368	107,136,68	0.03	0.02	0.02	0.07	0.03	0.02	0.02	nd	nd	nd
Trimethyl pyrazine	15.411	122,42,81	nd	nd	nd	nd	nd	nd	nd	nd	0.07	nd
Octanal	15.521	43,41,57	0.07	0.08	0.05	0.16	0.04	0.04	0.05	0.01	nd	0.35
<i>trans,trans</i> -2,4-Heptadienal	15.954	81,110,53	0.16	0.21	0.12	0.71	0.21	0.17	0.32	0.01	nd	0.08
N,N-dimethylmethanethioamide	16.624	89,44,42	nd	nd	nd	0.02	nd	nd	nd	0.08	0.10	nd
Unknown amine	16.686	58,44,42	nd	nd	nd	nd	nd	nd	nd	nd	0.60	nd
Phenylacetaldehyde	17.911	91,92,120	2.14	0.27	0.09	0.24	0.16	0.08	0.43	0.12	nd	2.71
3,4-Dimethyl-2,5-hexanedione	18.222	43,131,88	nd	nd	nd	nd	nd	nd	nd	0.14	nd	nd
Unknown	18.362	58,42,72	nd	nd	nd	nd	nd	nd	nd	nd	0.11	nd
4,7-Dimethyl-undecane	18.959	57,43,71	0.04	0.06	0.04	0.15	0.07	0.05	nd	0.02	nd	0.13
4-Methoxy-2,5-dimethyl-3(2H)-furanone	19.240	142,43,71	nd	nd	nd	nd	nd	nd	nd	nd	nd	0.09
<i>cis</i> -5-Nonenal	20.368	79,41,55	0.11	nd	nd	nd	nd	nd	nd	nd	nd	nd
1,2,4-Trithiolane	20.587	124,78,45	nd	nd	nd	nd	nd	nd	nd	nd	0.23	0.07
N,N,N'-Trimethyl-1,2-ethanediamine	21.221	58,42,59	nd	nd	nd	nd	nd	nd	nd	nd	0.45	nd
2-Nonanone	21.373	43,58,71	0.02	0.02	0.01	0.07	0.01	0.03	nd	nd	nd	0.10
Unknown aromatic ketone	21.666	107,79,135	nd	nd	nd	0.332	nd	0.13	nd	nd	nd	nd
Nonanal	22.105	57,41,56	0.14	0.41	0.46	0.42	0.14	0.10	0.12	0.12	nd	1.92
Dimethylaminohexane	24.123	58,45,42	nd	nd	nd	nd	nd	nd	1.03	nd	nd	nd
2,6,6-Trimethyl-2-cyclohexene-1,4-dione	24.568	68,96,152	nd	nd	nd	nd	nd	nd	nd	nd	nd	0.20
<i>cis,trans</i> -2,6-Nonadienal	25.324	41,70,69	0.05	0.06	0.04	0.22	0.04	0.05	0.06	nd	nd	0.04
2-Phenylpropanal	25.360	103,104,132	0.36	0.06	0.02	0.34	0.02	0.55	0.12	0.22	nd	nd
4-Ethyl benzaldehyde	25.702	104,133,105	0.07	0.07	0.03	0.24	0.06	0.07	0.07	nd	nd	nd
N,N-dimethyl-1-Butanamine	25.797	58,45,42	nd	nd	nd	nd	nd	nd	nd	nd	2.71	nd
5,6-Dihydro-5-methyl-4H-1,3,5-dithiazine	27.744	135,44,57	nd	nd	nd	nd	nd	nd	nd	nd	0.49	nd
Decanal	28.902	57,43,41	0.03	0.04	0.04	0.04	0.03	nd	0.03	nd	nd	0.11
Unknown amine	29.726	58,42,44	nd	nd	nd	nd	nd	nd	nd	nd	0.22	nd
Unkown hydrocarbon	37.834	57,42,177	1.39	0.31	0.34	nd	nd	nd	nd	nd	nd	nd
Pentadecane	38.260	57,43,71	nd	nd	nd	0.43	0.22	nd	nd	nd	nd	nd
Unknown (1,4 dimethyl piperidine)	39.160	112,57,42	nd	nd	nd	nd	nd	nd	nd	nd	0.14	nd
Unknown thiole	39.636	42,162,120	nd	nd	nd	nd	nd	nd	nd	nd	5.09	nd
Heptadecane	41.498	57,71,43	nd	0.04	nd	1.68	nd	0.15	nd	0.03	0.02	0.09
2,6,10,14-Tetramethyl pentadecane	41.598	57,71,43	0.06	0.04	nd	0.03	1.19	nd	0.15	0.03	nd	0.02
2-Pentadecanone	41.857	58,43,71	nd	nd	nd	nd	nd	nd	0.10	nd	nd	nd
<i>cis</i> -1-Octadecene	42.800	83,97,55	0.05	nd	0.08	0.10	0.05	0.07	0.10	0.03	0.02	0.11
Octadecane	43.070	57,71,43	0.26	0.71	nd	0.94	1.22	0.34	nd	0.07	0.20	nd
1-Nonadecene	45.049	55,83,97	nd	nd	0.07	nd	nd	0.06	0.09	0.07	nd	0.14
Octadecanal	45.409	57,43,82	nd	nd	nd	nd	nd	nd	nd	nd	nd	0.96
Octadecanoic acid ethyl ester	46.872	88,101,43	0.02	0.02	0.01	0.10	0.11	0.01	nd	0.01	0.01	0.02

\*: Only compounds present at concentrations &gt;0.1 µg/kg are included.

fects on some volatiles in other cases (Sérot *et al.*, 2001).

In this study, mussels exhibited the highest total concentration of volatiles, while among fish pilchard was the species with the highest number of volatile compounds and with the highest total concentrations (Table 4).

Fatty fish species like pilchard, anchovy, and picarel contained higher amounts of carbonyl compounds (12.5, 8.0 and 5.6 µg/Kg, respectively), when compared to the lean species bogue and hake (3.4 and 3.9 µg/Kg, respectively). The existing literature confirms the high lipid contents in sardines (Kalogeropoulos *et al.*, 2004; Zlatanov & Laskaridis, 2007), and anchovy and picarel, from the same origin as in our study, by the end of winter (Zlatanov & Laskaridis, 2007). The increased contents of lipid-originating carbonyl compounds in fatty fish when compared to lean fish has been reported previously (Iglesias *et al.*, 2010).

Distinct patterns of volatiles were observed among the different seafood. All finfish species were found to contain 2,3-pentanedione, which was absent from the shellfish studied, and up to 30 times higher 1-penten-3-ol concentrations. The low concentrations or absence of 1-penten-3-ol have been confirmed in mussels (Le Guen *et al.*, 2000a) and pre-cooked shrimp (Soncin *et al.*, 2008). The non-fish seafood was also characterized by the absence of *trans*-2-pentenal, *trans*-2-(2-pentenyl) furan,

and 4-ethyl-benzaldehyde. Finfish also contained much higher quantities of 2-ethyl furan compared to shellfish, in agreement with the findings of Giogios *et al.* (2009) who reported higher 2-ethylfuran and furans concentrations in fish oils when compared to shrimp oil. Among shellfish, only mussels contain significant quantities of this compound, which were in any case lower than that of fish species. The volatiles of pilchard contained the highest amounts of 2-ethylfuran, being more than 5 times higher than in the other species. A potential mechanism for 2-ethylfuran formation in fish muscle has been proposed, including n-3 fatty acid β-oxidation to produce conjugated dienes radicals, which are further oxidized to produce vinylhydroperoxides; these in turn lose a hydroxyl radical and undergo cyclization to form 2-ethylfuran (Medina *et al.*, 1999).

Pilchard volatiles are characterized by the presence of significant amounts of alcohols when compared to the other fish and seafood species (Table 4), with 1-penten-3-ol, 1-methoxy-2-propanol, *trans*-2-penten-1-ol and 3-pentanol predominating (Table 3). These alcohols have been confirmed previously in sardines of various origins (Prost *et al.*, 2004; Ganeko *et al.*, 2008). With respect to the other pilchard volatiles, the profound presence of various carbonyls (2,3-pentanedione, pentanal, 2-butanone, *cis*-2-pentenal, hexanal, *trans,trans*-2,4-heptadienal and

**Table 4.** Concentrations of volatile compound classes (µg/kg fresh weight) characterizing the studied Mediterranean seafood species.

	<i>A. boyeri</i>	<i>S. smaris</i>	<i>M. merluccius</i>	<i>S. pilchardus</i>	<i>B. boops</i>	<i>E. encrasicolus</i>	<i>M. barbatus</i>	<i>L. vulgaris</i>	<i>P. longirostris</i>	<i>M. galloprovincialis</i>
Alcohols	1.75	1.71	0.69	8.05	0.77	0.82	1.15	0.94	0.12	0.64
Aldehydes	9.27	3.09	2.32	9.68	2.54	4.10	4.47	2.04	0.70	19.47
Ketones	1.58	2.54	1.62	2.78	0.86	3.93	2.17	2.17	0.85	2.19
Furans	1.29	0.53	0.40	2.78	0.64	0.68	0.74	0.19	0.09	3.55
Amines	0.01	0.01	0.01	0.02	tr <sup>a</sup>	0.02	0.00	0.45	0.92	nd <sup>b</sup>
Pyrazines	nd	nd	tr	nd	0.12	0.09	0.11	0.18	1.71	1.64
Pyridines	nd	0.27	0.01	0.01	0.05	nd	0.02	0.03	0.06	1.54
Pyrroles	nd	nd	nd	nd	nd	nd	tr	0.01	0.11	0.42
Hydrocarbons	2.06	2.17	1.26	17.09	3.97	4.17	8.15	5.08	0.83	30.14
Sulfur compounds	0.27	0.06	0.04	0.24	0.05	0.12	0.05	0.27	0.90	0.36
Ethers	0.26	0.21	0.21	0.23	0.05	0.07	0.12	0.12	0.07	0.24
Esters	0.12	0.10	0.07	0.22	0.31	0.14	0.10	0.01	0.13	0.10
Acids	0.01	nd	tr	nd	nd	0.01	nd	nd	nd	nd
Others	0.07	0.06	0.04	0.09	0.03	0.14	0.02	0.10	0.05	0.05
Unknown	2.00	0.51	0.47	0.85	0.49	0.84	1.34	2.53	11.5	1.50
TOTAL	18.7	11.22	7.17	42.04	9.79	15.06	18.38	17.07	16.54	60.35

tr stands for "traces"; nd stands for "not detected".

*trans*-4-heptenal) and 2-ethylfuran in sardines have been reported previously and have been considered to contribute to a characteristic sardine-fishy odour (Triqui & Bouchriti, 2003; Prost *et al.*, 2004; Ganeko *et al.*, 2008).

To the best of our knowledge, there are no published data for the volatile compounds of the other fish species of this study, with the exception of carbonyls in the European anchovy (Iglesias *et al.*, 2010).

Squid was herein characterized by high concentrations of 1-methoxy-2-propanol, 3-hydroxy-butanone, an unknown compound (retention time  $R_t=12.043$ min), N,N-dimethyl methanethioamide, 2,4-dihydroxy-3,4-dimethyl-2,5-hexanedione, and also by the very low furan content and the almost complete absence of esters. No previous data exist regarding the volatile compounds of fresh squid. The only available data refer to processed (Giri *et al.*, 2011) or cooked (Kubota *et al.*, 1996) squid, or to offensive volatile compounds from decaying squid (Kim *et al.*, 2009).

The shrimp is characterized by the presence of amines and S-compounds, low concentrations of hydrocarbons, alcohols and carbonyls and by the almost complete absence of furans. Mussels, on the contrary, are characterized by high contents of hydrocarbons, aldehydes, furans and furan-containing compounds. What shrimp and mussels have in common is the presence of S-compounds and N-containing heterocyclic compounds (mainly pyrazines).

Among the S-compounds found in shrimp, the most abundant is an unknown thiol (Table 3). The S-containing compounds, are considered to be enzymatically produced in seafood from methionine and cysteine (Varlet & Fernandez, 2010). Thioamides, herein present in shrimp, have also been reported for crab (Chung, 1999) and dried squid (Kawai *et al.*, 1991). Thiazoles, have been found in cuttlefish oil (Shen *et al.*, 2007) and cooked squid (Kubota *et al.*, 1996) and have been reported as heating products in seafood (Varlet & Fernandez, 2010).

The N-containing heterocyclic compounds have been confirmed for various crab species (Chung 1999; Chen & Zhang, 2006), and they have been positively related to prawn-shrimp aroma notes (Morita *et al.*, 2001).

Previous studies on mussels have also indicated the existence of pyrazines and pyridine-containing compounds (Le Guen *et al.*, 2000a), and shown the presence of significant amounts of carbonyls, in particular hexanal, benzaldehyde, and 2-heptanone (Le Guen *et al.*, 2000a; Fuentes *et al.*, 2009).

On the other hand, some differentiations were found in relation to literature data. *Trans*-4-heptenal, which has been found in great quantities (Fuentes *et al.*, 2009) and recognized as one of the most potent odorants in mussel (Le Guen *et al.*, 2000b), is present in relatively small quantities in our mussels. Also 2,3-diones, indicated by Le Guen *et al.* (2000a, 2000b) as major mussel odorants, are almost absent in our mussels but present in the fish species. The same also applies for 2,4-octadienal, that

was absent from mussels in our study, but has been reported as being a major mussel odorant contributor (Le Guen *et al.*, 2000b). On the other hand, the presence of significant quantities of *cis*-2-butenal, 2-methyl- and 3-methyl- butanal, pentanal, 1-(2-furanyl)-ethanone and 5-methyl-2-furancarboxaldehyde has not been reported previously in the respective literature.

Although no profound justification occurs for these differences, possible explanations can be found. Given that the volatile compounds' isolation methodology followed in the literature (Le Guen *et al.*, 2000b; Fuentes *et al.*, 2009), was the same as ours, these differences can only be justified by the origin of the samples. Mussels, being filter feeders, may be prone to origin-specific environmental impacts on their volatile compounds (Fuentes *et al.*, 2009). Another possible explanation for the observed differences between literature data and our results can be the different sampling season, since seasonality in seafood volatile compounds is known to occur (Ólafsdóttir & Jónsdóttir, 2010). In the works of Le Guen *et al.* (2000a) and Fuentes *et al.* (2009), sampling was conducted during a different period of the year than this study, while in other cases the sampling season is not specified (Le Guen *et al.*, 2000b).

Cluster analysis showed that the volatiles pattern of pilchard primarily, and mussel to a lesser degree, followed by shrimp, differentiate them from the rest of the studied seafood (Fig. 1). These three species also exhibited the highest squared Euclidian distances (6216.6 and 6016.4 between pilchard, and mussel and shrimp respectively). Four fish species, namely hake, mullet, picarel and bogue, are grouped together (low squared Euclidian distances, ranging from 74.9 between mullet and hake to 345.4 between picarel and bogue). The squid and sand smelt differentiate from this group, while anchovy exhibits even higher differentiation.

What is interesting in this clustering as regards the fish species is that the two littoral species inhabiting very shallow waters, i.e. the pilchard and anchovy differ as regards their volatile compounds from the rest of the fish species that are pelagic, inhabiting deeper waters, demersal or benthic ([www.fishbase.org](http://www.fishbase.org)). The fact that all studied fish are carnivorous ([www.fishbase.org](http://www.fishbase.org)) and that they originate from the same geographic region, indicates that the ecology of the fish species plays an important role in their volatile compounds profile.

Principal components analysis (PCA) for individual volatile compounds extracted nine components explaining almost 100 % of total variance in seafood species (Fig. 2). The contribution of the volatiles to each component is detailed in Appendix Table 1. In the first component, explaining 34.4% of total variability, numerous carbonyls are present, including aliphatic, non aliphatic and heterocyclic ones, three carbonyl-furans and one S-containing aldehyde. Also, two alcohols (1-pentanol, 3-pentanol), one furan (tetrahydrofuran), numerous



\* \* \* H I E R A R C H I C A L C L U S T E R A N A L Y S I S \* \* \*

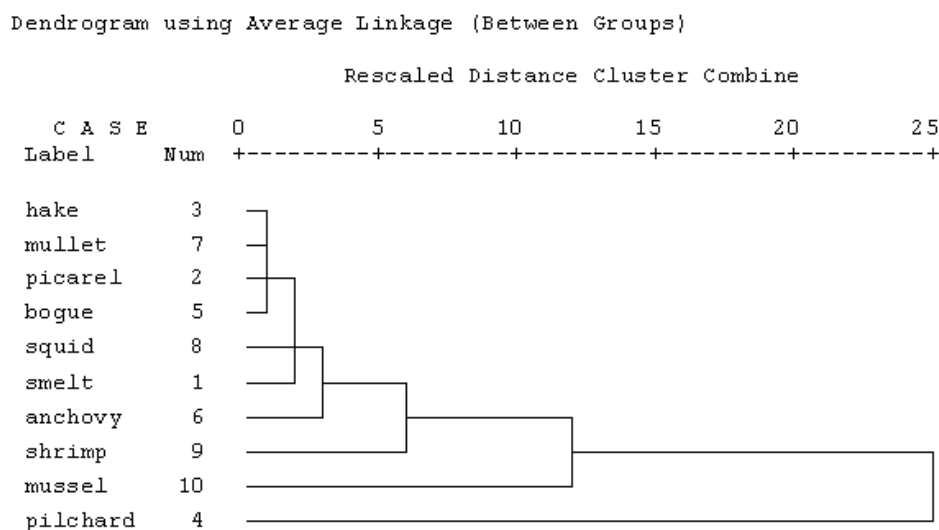


Fig. 1: Cluster analysis: dendrogram using average linkage between seafood species based on their volatile compound profiles.

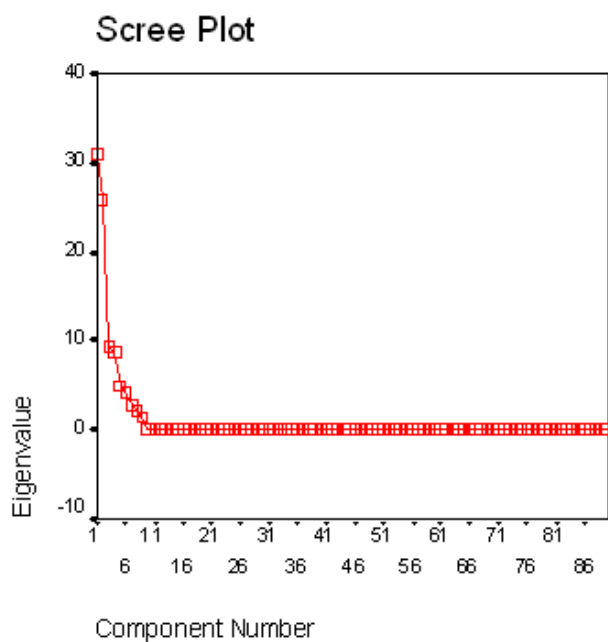


Fig. 2: Scree plot obtained from PCA of volatile compounds of the studied species.

saturated, unsaturated, aromatic hydrocarbons, and one S-containing (1-methylthiopropene), a terpene (d-limonene), and one N-containing compound (pyridine) are present in the 1<sup>st</sup> component. The second component, explaining, cumulatively with the first one, 63.2% of total variability includes non aliphatic and cyclic carbonyls, four furans, two alcohols (1-penten-3-ol, *trans*-2-penten-1-ol), six non aliphatic and aromatic hydrocarbons, one ether (2-methoxy-2-methyl-butane), one ester (hexadecanoic acid-ethylester), and one S-compound (dimethyl-trisulfide). And the third component that, cumulatively with

the other two, explains 73.4% of total variability contains three esters, two cyclic carbonyls (acetophenone and thiophenecarboxaldehyde), one alcohol (1-octadecanol) and eight hydrocarbons. The contribution of carbonyls of six, eight and nine carbon atoms, in the three first components underlines the importance of those substances. They are mostly lipoxygenase-generated (Ólafsdóttir & Jónsdóttir, 2010), considered typical of fresh seafood and have been associated with aroma notes characterized as light, green, delicate, marine plant and iodised (Josephson *et al.*, 1984; Lindsay, 1990; Ólafsdóttir & Jónsdóttir, 2010)

## Conclusions

Besides the common seafood-characterizing carbonyls, some of the studied fish-species have distinct volatile compound patterns. Thus, the pilchard is characterized by a higher number and content of volatiles and in particular alcohols. The non-fish species also differ, with mussel showing the most distinct pattern of volatiles, characterized by high contents of aldehydes, furans and furan-containing compounds, and N-containing heterocyclic compounds (pyridines, pyrazines and pyrroles), and shrimp being rich in amines and S-compounds. An interesting observation is that the clustering of fish species according to their volatile compound profiles may be related to their ecological inhabitants.

## Acknowledgement

This work was funded by the Greek General Secretariat for Research and Technology, Ministry of Education, Life Long Learning and Religious Affairs, CO-

OPERATION 2011-Partnerships of Production and Research Institutions in Focused Research and Technology Sectors. Program: "KRANIOS".

## References

- Alasalvar, C., Taylor, K.D.A., Shahidi, F., 2005. Comparison of volatiles of cultured and wild sea bream (*Sparus aurata*) during storage in ice by dynamic headspace analysis/gas chromatography-mass spectroscopy. *Journal of Agricultural and Food Chemistry*, 53, 2616-2622.
- Chen, D., Zhang, M., 2006. Analysis of volatile compounds in Chinese mitten crab (*Eriocheir Sinensis*). *Journal of Food and Drug Analysis*, 14, 297-303.
- Chung, H.Y., 1999. Volatile components of *Charybdis feriatius*. *Journal of Agricultural and Food Chemistry*, 47, 2280-2287.
- Fuentes, A., Fernández-Segovia, I., Escriche, I., Serra, J.A., 2009. Comparison of physico-chemical parameters and composition of mussels (*Mytilus galloprovincialis* Lmk.) from different Spanish origins. *Food Chemistry*, 112, 295-302.
- Ganeko, N., Shoda, M., Hirohara, I., Bhadra, A., Ishida, T. et al., 2008. Analysis of volatile flavor compounds of sardine (*Sardinops melanostica*) by solid phase microextraction. *Journal of Food Science*, 73, S83-S88.
- Giogios, I., Grigorakis, K., Nengas, I., Papisolomondos, S., Papaioannou et al., 2009. Fatty acid composition and volatile compounds of selected marine oils and meals. *Journal of the Science of Food and Agriculture*, 89, 88-100.
- Giri, A., Osako, K., Ohshime, T., 2011. Effects of hypobaric and temperature-dependent storage on headspace aroma-active volatiles in common squid miso. *Food Research International*, 44, 739-747.
- Grigorakis, K., Taylor, K.D.A., Alexis, M.N., 2003. Organoleptic and volatile aroma compounds comparison of wild and cultured gilthead sea bream: sensory differences and possible chemical basis. *Aquaculture*, 225, 109-119.
- Grigorakis, K., Fountoulaki, E., Giogios, I., Alexis, M.N., 2009. Volatile compounds and organoleptic qualities of gilthead sea bream (*Sparus aurata*) fed commercial diets containing different lipid sources. *Aquaculture*, 290, 116-121.
- Howgate, P., Johnston, A., Whittle, K.J., 1992. *Multilingual guide to EC freshness grades for fishery products*. Food Safety Directorate, Ministry of Agriculture, Fisheries and Food, Torrey Research Station, Aberdeen, Scotland, pp. 32.
- Iglesias, J., Gallardo, J.L., Medina, I., 2010. Determination of carbonyl compounds in fish species samples with solid-phase microextraction with on-fibre derivatization. *Food Chemistry*, 123, 771-778.
- Josephson, D.B., Lindsay, R.C., Stuibler, D.A., 1984. Variations in the occurrences of enzymically derived volatile aroma compounds in salt- and freshwater fish. *Journal of Agricultural and Food Chemistry*, 32, 1344-1347.
- Kalogeropoulos, N., Andrikopoulos, N.K., Hassapidou, M., 2004. Dietary evaluation of Mediterranean fish and molluscs pan-fried in virgin olive oil. *Journal of the Science of Food and Agriculture*, 84, 1750-1758.
- Kawai, T., 1996. Fish flavor. *Critical Reviews in Food Science and Nutrition*, 36, 257-298.
- Kawai, T., Ishida, Y., Kakiuchi, H., Ikeda, N., Higashida, T. et al., 1991. Flavour components of dried squid. *Journal of Agricultural and Food Chemistry*, 39, 770-777.
- Kim, K.-H., Pal, R., Ahn, J.-W., Kim, Y.-H., 2009. Food decay and offensive odorants: A comparative analysis among three types of food. *Waste Management*, 29, 1265-1273.
- Kubota, K., Matsukage, Y., Sekiwa, Y., Kobayashi, A., 1996. Identification of the characteristic volatile flavor compounds formed by cooking squid (*Todarodes pacificus* Steenstrup). *Food Science & Technology International*, 2, 163-166.
- Le Guen, S., Prost, C., Demaimay, M., 2000a. Characterization of odorant compounds of mussels (*Mytilus edulis*) according to their origin using gas chromatography-olfactometry and gas chromatography-mass spectrometry. *Journal of Chromatography A*, 896, 361-371.
- Le Guen, S., Prost, C., Demaimay, M., 2000b. Critical comparison of three olfactometric methods for the identification of the most potent odorants in cooked mussels (*Mytilus edulis*). *Journal of Agricultural & Food Chemistry*, 48, 1307-1314.
- Leduc, F., Tournayre, P., Kondjoyan, N., Mercier, F., Malle, P. et al., 2012. Evolution of volatile odorous compounds during the storage of European seabass (*Dicentrarchus labrax*). *Food Chemistry*, 131, 1304-1311.
- Lindsay, R.C., 1990. Fish Flavors. *Food Reviews International*, 6, 437-455.
- Medina, I., Satue-Gracia, M.T., Frankel, E.N., 1999. Static headspace-gas chromatography analysis to determinate oxidation of fish muscle during thermal processing. *Journal of the American Oil Chemists' Society*, 76, 231-236.
- Morita, K., Kubota, K., Aishima, T., 2001. Sensory characteristics and volatile components in aromas of boiled prawns prepared according to experimental designs. *Food Research International*, 34, 473-481.
- Morita, K., Kubota, K., Aishima, T., 2003. Comparison of aroma characteristics of 16 fish species by sensory evaluation and gas chromatographic analysis. *Journal of the Science of Food and Agriculture*, 83, 289-97.
- Ólafsdóttir, G., Jónsdóttir, R., 2010. *Volatile aroma compounds in fish* p. 97-117. In: *Handbook of seafood and seafood product analysis*. Nollet, L.M.L., Toldrá, F. (Eds). Boca Raton, FL: CRC Press.
- Pennarun, A.L., Prost, C., Demaimay, M., 2002. Identification and origin of the character-impact compounds of raw oyster *Crassostrea gigas*. *Journal of the Science of Food and Agriculture*, 82, 1652-1660.
- Piveteau, F., Le Guen, S., Gandemer, G., Baud, J.B., Prost, C. et al., 2000. Aroma of fresh oysters *Crassostrea gigas*: composition and aroma notes. *Journal of Agricultural & Food Chemistry*, 48, 4851-4857.
- Prell, P.A., Sawyer, F.M., 1988. Flavor profiles of 17 species of north Atlantic fish. *Journal of Food Science*, 53, 1036-1042.
- Prost, C., Hallier, A., Cardinal, M., Serot, T., Courcoux, P., 2004. Effect of storage time on raw sardine (*Sardina pilchardus*) flavor and aroma quality. *Journal of Food Science*, 69, S198-S204.
- Rasmussen, T., Anthoni, U., Christophersen, C., Nielsen, P.H., 1993. Volatile compounds for marine indicator organism *Mytilus edulis*. *Chemosphere*, 27, 2123-2125.
- Sérot, T., Regost, C., Prost, C., Robin, J., Arzel, J., 2001. Effect of dietary lipid sources on odour-active compounds in muscle of turbot (*Psetta maxima*). *Journal of the Science of Food and Agriculture*, 81, 1339-1346.
- Shen, S., Xie, J., Xu, X., 2007. The components of cuttlefish (*Sepiella maindroni* de Rochebruns) oil. *Food Chemistry*, 102, 210-214.

- Silva, J.M.G., Valente, L.M.P., Castro-Cunha, M., Bacelar, M., Guedes De Pinho, P., 2012. Impact of dietary plant protein levels on the volatile composition of Senegalese sole (*Solea senegalensis* Kaup, 1858) muscle. *Food Chemistry*, 131, 596-602.
- Sonsin, S., Chiesa, L.M., Panseri, S., Biondi, P., Cantoni, C., 2008. Determination of volatile compounds of precooked prawn (*Penaeus vannamei*) and cultured gilthead sea bream (*Sparus aurata*) stored in ice as possible spoilage markers using solid phase microextraction and gas chromatography/mass spectrometry. *Journal of the Science of Food and Agriculture*, 89, 436-442.
- Triqui, R., Bouchriti, N., 2003. Freshness assessments of Moroccan sardine (*Sardina pilchardus*): Comparison of overall sensory changes to instrumentally determined volatiles. *Journal of Agricultural and Food Chemistry*, 51, 7540-7546.
- Varlet, V., Fernandez, X., 2010. Sulfur-containing volatile compounds in seafood: occurrence, odorant properties and mechanisms of formation. *Food Science and Technology International*, 16, 463-503.
- Zhang, Z., Li, G., Luo, L., Chen, G., 2010. Study on seafood volatile profile characteristics during storage and its potential use for freshness evaluation by headspace solid phase microextraction coupled with gas chromatography-mass spectrometry. *Analytica Chimica Acta*, 659, 151-158.
- Zlatanov, S., Laskaridis, K., 2007. Seasonal variation in the fatty acid composition of three Mediterranean fish – sardine (*Sardina pilchardus*), anchovy (*Engraulis encrasicolus*) and picarel (*Spicara smaris*). *Food Chemistry*, 103, 725-728.